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# Suzuki Ignis Rg413 2002 2006 Workshop Manual

**Bob Henderson, A. K. Legg** 

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web figure 5 5 1 a potential energy diagram shows the total potential energy of a reacting system as the reaction proceeds a in an endothermic reaction the energy of the products is greater than the energy of the reactants and  $\Delta$  h is positive b in an exothermic reaction the energy of the products is lower than the energy of the

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k e ea rt where k depends by many factors as solvent geiometry type of reaction ea is the activation energy r is the universal costant of gases t is the absolute temperature

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web the energy of the reactants is shown on the energy diagram above on the right side of the energy diagram draw a horizontal line segment to indicate the energy of the products draw a vertical line segment to indicate  $\Delta h$  for the reaction and label it with the correct value see attached image

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