# Covalent Bonding Worksheet

Covalent bonding occurs when two or more NON METALS share electrons, attempting to attain a stable octet (8 outer electrons)in their outer shell for at least part of the time. Draw a Lewis dot diagram for each element listed. Circle the unpaired electrons that will be shared between the elements.

H<sub>2 brothrogen</sub> is distornic

H + H → H → H-H Single Bond

2.) F2 thurston in discounts

3.) O2 oxygen is diatomic

O + O

 $\bullet$   $\bigcirc =\bigcirc$ 

Double Bond

4.) No nitrogen is diatomic. Is this a triple bond?

N + N

5.) BF<sub>3 you need 3 fluorine atoms here</sub>

 $\mathbf{R} + \mathbf{F}$ 

6.) Ammonia NH3 hist: how many hydrogen atoms are needed?

N + H

7.) Carbon dioxide CO<sub>2</sub>

0 + c + 0

Methane CH<sub>4</sub> careful here 4 hydrogen atoms needed

C + H

9.) Dihydrogen monoxide: the most dangerous substance on the planet. It has killed more people than any other substance known to mankind!!

H + O + H

10.) SO<sub>2</sub> hint: one pair of electrons from sulfur must be slit up for this one to work.

O + S + O

# **The Nature Of Covalent Bonding Worksheet Answers**

**EW Minium** 

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