

# CHAPTER 15 STUDY GUIDE

## Energy and Chemical Change

### Section 15.1 Energy

In your textbook, read about the nature of energy.

In the space at the left, write *true* if the statement is true; if the statement is false, change the italicized word or phrase to make it true.

- \_\_\_\_\_ 1. *Energy* is the ability to do work or produce heat.
- \_\_\_\_\_ 2. The law of conservation of energy states that energy *can be* created and destroyed.
- \_\_\_\_\_ 3. Chemical potential energy is energy stored in a substance because of its *composition*.
- \_\_\_\_\_ 4. *Heat* is a form of energy that flows from a warmer object to a cooler object.
- \_\_\_\_\_ 5. A calorie is the amount of energy required to raise the temperature of *one gram* of pure water by one degree Celsius.
- \_\_\_\_\_ 6. A *calorie* is the SI unit of heat and energy.
- \_\_\_\_\_ 7. The *specific* heat of a substance is the amount of heat required to raise the temperature of one gram of that substance by one degree Celsius.
- \_\_\_\_\_ 8. *Kinetic energy* is energy of motion.
- \_\_\_\_\_ 9. Chemicals participating in a chemical reaction contain only *potential energy*.
- \_\_\_\_\_ 10. One nutritional Calorie is equal to *100 calories*.
- \_\_\_\_\_ 11. One calorie equals *4.184 joules*.
- \_\_\_\_\_ 12. When a fuel is burned, some of its *chemical potential energy* is lost as heat.
- \_\_\_\_\_ 13. To convert kilojoules to joules, *divide* the number of kilojoules by 1000 joules/1 kilojoule.

Answer the following question. Show all your work.

14. If the temperature of a 500.0-g sample of liquid water is raised 2.00°C, how much heat is absorbed by the water? The specific heat of liquid water is 4.184 J/(g·°C).

# Standardized Test Practice Chapter 15 Chemistry Answers

**Ralph H. Petrucci, Robert K. Wismer**



## **Standardized Test Practice Chapter 15 Chemistry Answers:**

*Cracking the SAT Chemistry Subject Test* Princeton Review, 2014-12-09 EVERYTHING YOU NEED TO HELP SCORE A PERFECT 800 Equip yourself to ace the SAT Chemistry Subject Test with The Princeton Review's comprehensive study guide including 3 full length practice tests thorough reviews of key chemistry topics and targeted strategies for every question type We don't have to tell you how tough SAT Chemistry is or how helpful a stellar exam score can be for your chances of getting into your top choice college Written by the experts at The Princeton Review *Cracking the SAT Chemistry Subject Test* arms you to take on the test and achieve your highest score Techniques That Actually Work Tried and true strategies to help you avoid traps and beat the test Tips for pacing yourself and guessing logically Essential tactics to help you work smarter not harder Everything You Need to Know for a High Score Expert subject reviews for every test topic Up to date information on the SAT Chemistry Subject Test Score conversion tables for accurate self assessment Practice Your Way to Perfection 3 full length practice tests with detailed answer explanations Hands on experience with all three question types in each content chapter Complete study sheet of core formulas and terms [A Level Chemistry MCQ \(Multiple Choice Questions\)](#) Arshad Iqbal, 2019-06-18 The A Level Chemistry Multiple Choice Questions MCQ Quiz with Answers PDF A Level Chemistry MCQ PDF Download Quiz Questions Chapter 1 28 Practice Tests with Answer Key IGCSE GCE Chemistry Questions Bank MCQs Notes includes revision guide for problem solving with hundreds of solved MCQs A Level Chemistry MCQ with Answers PDF book covers basic concepts analytical and practical assessment tests A Level Chemistry MCQ PDF book helps to practice test questions from exam prep notes The A Level Chemistry MCQs with Answers PDF eBook includes revision guide with verbal quantitative and analytical past papers solved MCQs A Level Chemistry Multiple Choice Questions and Answers MCQs PDF Free download chapter 1 a book covers solved quiz questions and answers on chapters Alcohols and esters atomic structure and theory benzene chemical compound carbonyl compounds carboxylic acids acyl compounds chemical bonding chemistry of life electrode potential electrons in atoms enthalpy change equilibrium group IV groups II and VII halogenoalkanes hydrocarbons introduction to organic chemistry ionic equilibria lattice energy moles and equations nitrogen and sulfur organic and nitrogen compounds periodicity polymerization rates of reaction reaction kinetics redox reactions and electrolysis states of matter transition elements tests for college and university revision guide A Level Chemistry Quiz Questions and Answers PDF free download eBook's sample covers beginner's solved questions textbook's study notes to practice online tests The book IGCSE GCE Chemistry MCQs Chapter 1 28 PDF includes high school question papers to review practice tests for exams A Level Chemistry Multiple Choice Questions MCQ with Answers PDF digital edition eBook a study guide with textbook chapters tests for IGCSE NEET MCAT GRE GMAT SAT ACT competitive exam A Level Chemistry Mock Tests Chapter 1 28 eBook covers problem solving exam tests from chemistry textbook and practical eBook chapter wise as Chapter 1 Alcohols and Esters MCQ Chapter 2 Atomic Structure and Theory MCQ Chapter 3 Benzene Chemical Compound

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Cracking the AP Chemistry Exam, 2013 Edition Paul Foglino, Princeton Review, 2012-08-07 Provides techniques for achieving high scores on the AP chemistry exam and includes two full length practice tests a subject review for all topics and sample questions and answers

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In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Standardized Test Practice Chapter 15 Chemistry Answers**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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