

Exploring States of Matter

Virtual lab simulation of particle behavior at different states of matter in various substances.
Directions:

- A. Open the link to explore the states of matter simulation. Click on "STATES"
https://phet.colorado.edu/en/simulations/states-of-matter/!states-of-matter_en.html
- B. Find the thermometer, heating and cooling toggle, 4 different atoms/molecules, and 3 different states of matter.
- C. Carry out the following activities to explore how the particles in different states of matter behave at different temperatures.

Before you begin:

- A. **Look around you** find something near you at the 3 different states of matter.
 - a. Solid []
 - b. Liquid []
 - c. Gas []
- B. **Think about it and explain:** How is it possible for things to be at different states of matter even though they are at the same temperature?
 []
- C. **Practice using the cooling and heating toggles:** Are you comfortable with using the toggle to change the temperature of the substance in the container? (Don't move on until you can answer "yes" to this question.)
 []

Activity 1: Exploring the Melting Point and Boiling Point of Oxygen:

1. **Background Research** Use Google to find the answers to these 4 questions:
 - a. What is regular room temperature in Celsius?
 []
 - b. What is the melting point of Oxygen?
 []
 - c. What is the boiling point of Oxygen?
 []
 - d. What state of matter is Oxygen at room temperature?
 []
2. **Click on Oxygen and Solid:** We are now looking at solid oxygen.
 - a. What temperature is the thermometer at?
 []
 - b. Describe how the particles are arranged and behaving in solid Oxygen.
 []

States Of Matter Virtual Lab Answers

University of Illinois (System)

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the University of Illinois (System) text.

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