# F324-unofficial mark scheme (in progress)

# Marks accounted for: 60/60

- Give the systematic name of the alcohol that is formed by this hydrolysis. (1)
  - Propan(e)-1,2,3-triol
- What are the other monomers that make up this polymer? (3)

(Hydrolysed by hot NaOH)

- Two different carboxylic acids with COO'(Na") The monomers have COOH
  as it's not referring to the hyrolysed products but the monomer units of
  the polymer I'm pretty sure that the question said: "What are the other
  products of this hydrolysis" or something along those lines
- One salt had a c=c double bond
- Why are people who drink a lot of goat's milk likely to suffer from CHD? (2)
  - Fatty acids build up in the arteries and lead to CHD?
  - Has trans isomer due to c=c double bond fatty acid linked to LDL's and hence Atherosclerosis
  - Trans-fatty acid in triglyceride of milk increases 'bad' cholesterol levels in blood, resulting in increased risk of coronary heart disease and stroke (straight from the spec)
  - trans fatty acids can pack more tightly and hence have higher melting points than cis fatty acids. --> so more likely to deposit on artery walls
  - The other two molecules were fully saturated fatty acids which would also lead to the formation of atheroma
- State how 4-aminophenol can behave as a base. (1)
- The lone pair on the Nitrogen atom in the amine group can accept a proton, forming a dative covalent bond with it to give NH3+.
- NH2 + H+ ---> NH3+
- 4-Aminophenol is produced by the reduction of 4-nitrophenol. Write an equation to show the production of 4-aminophenol from 4-nitrophenol. Use [H] to represent the reducing agent. (1)
- 4-Nitrophenol + 6[H] = 4 Aminophenol + 2H20
- 4-nitrophenol can be produced from 4-bromophenol. (4)
  - Complete the mechanism for this reaction.
  - Use "NO, as the electrophile. Include any intermediate and the products.
  - In the mechanism for this reaction, NO<sub>2</sub> substitutes for Br on the ring.

# **Unofficial Mark Scheme Stats Mei 2013 June**

**VM** Jensen

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#### Table of Contents Unofficial Mark Scheme Stats Mei 2013 June

- 1. Understanding the eBook Unofficial Mark Scheme Stats Mei 2013 June
  - o The Rise of Digital Reading Unofficial Mark Scheme Stats Mei 2013 June
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Unofficial Mark Scheme Stats Mei 2013 June
  - Exploring Different Genres
  - o Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Unofficial Mark Scheme Stats Mei 2013 June
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Unofficial Mark Scheme Stats Mei 2013 June
  - Personalized Recommendations
  - Unofficial Mark Scheme Stats Mei 2013 June User Reviews and Ratings
  - Unofficial Mark Scheme Stats Mei 2013 June and Bestseller Lists
- 5. Accessing Unofficial Mark Scheme Stats Mei 2013 June Free and Paid eBooks
  - Unofficial Mark Scheme Stats Mei 2013 June Public Domain eBooks
  - Unofficial Mark Scheme Stats Mei 2013 June eBook Subscription Services
  - Unofficial Mark Scheme Stats Mei 2013 June Budget-Friendly Options
- 6. Navigating Unofficial Mark Scheme Stats Mei 2013 June eBook Formats
  - o ePub, PDF, MOBI, and More
  - Unofficial Mark Scheme Stats Mei 2013 June Compatibility with Devices
  - Unofficial Mark Scheme Stats Mei 2013 June Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - o Adjustable Fonts and Text Sizes of Unofficial Mark Scheme Stats Mei 2013 June
  - Highlighting and Note-Taking Unofficial Mark Scheme Stats Mei 2013 June
  - o Interactive Elements Unofficial Mark Scheme Stats Mei 2013 June
- 8. Staying Engaged with Unofficial Mark Scheme Stats Mei 2013 June

- o Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Unofficial Mark Scheme Stats Mei 2013 June
- 9. Balancing eBooks and Physical Books Unofficial Mark Scheme Stats Mei 2013 June
  - Benefits of a Digital Library
  - o Creating a Diverse Reading Collection Unofficial Mark Scheme Stats Mei 2013 June
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Unofficial Mark Scheme Stats Mei 2013 June
  - Setting Reading Goals Unofficial Mark Scheme Stats Mei 2013 June
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Unofficial Mark Scheme Stats Mei 2013 June
  - o Fact-Checking eBook Content of Unofficial Mark Scheme Stats Mei 2013 June
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - o Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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