

List of Madarsa Programs approved by the University for Admissions 2019-20

S. No.	Name of Madarsa	Madarsa courses	Recommended for the purpose of admission in MANUU programs
ANDHRA PRADESH			
1.	Jamiatus Salihat, Kadapa	Alimiat	B.A., B.A. Hons-JMC
2.	Jamiatun Nisran As Salafia 7-29, Thonawada, Chandragiri, Tirati, Chittour	Fazilat	B.A., B.A. Hons-JMC
3.	Jamia Mohammadia Arabia, Kanekal Road, Rayadung, Anantpur	Fazil	MA (Islamic Studies, Arabic, Urdu, Persian)
ASSAM			
4.	State Madrasa Education Board Assam Kahilipara, Guwahati	Fadilul- Ma' Arif (F.M.)	B.A., B.A. (Hons-JMC)
		Mumtazul-Muhaddithin	MA (Islamic Studies, Arabic, Urdu)
5.	Madrasahatul Banat & Banat Islamic Academy, Mahansari Pathar, Dist. Morigaon	Alimiat	B.A., B.A. (Hons-JMC)
6.	Jamia Islamia Jalalia, Gopalnagar, Hojai, Nagaon	Fazilat	MA (Islamic Studies, Arabic, Urdu, Persian)
BIHAR			
7.	Bihar State Madarsa Education Board	Moulvi	B.A., B.A. (Hons-JMC)
		Alimiat	M.A (Islamic Studies, Arabic, Urdu, Persian)
8.	Madrasah Shamsul Huda, Patna	Fazil	M.A (Islamic Studies, Arabic, Urdu, Persian)
9.	Madarsa Jama'atul Qadim, Darul Uloom-ul-Islamia, Pratap Ganj, Supaul	Alim	B.A., B.A. (Hons-JMC)
10.	Jamia ibne taimiya, Champaran	Alimiat	B.A., B.A. (Hons-JMC)
11.	Tauheed Education Trust, Kishan Ganj	Alim	B.A., B.A. (Hons-JMC)
12.	Darul Uloom Ahmadiya Salafia, Darbhanga	Alimiat	B.A., B.A. (Hons-JMC)
13.	Al-Madrasahatul Islamia, Raghunagar, Bhavara, Madhubani	Fazilat	B.A., B.A. (Hons-JMC)
14.	Jamiatul Banat, Gaya	Alimiat	B.A., B.A. (Hons-JMC)
15.	Madrasa Imanat-e-Sharia, Phoolwari Sahrif, Patna	Alimiat	B.A., B.A. (Hons-JMC)
16.	Darul Uloom Khairia Nizamia Mohalla Baradari Sahasram, Rohtas	Alimiat	B.A., B.A. (Hons-JMC)
17.	Jamia Rahmani Khanqah Makhsuspur, Munger	Alimiat	B.A., B.A. (Hons-JMC)
18.	Madrasa Islamia Muhiul Uloom Shakel Toli, Siwan	Alimiat	B.A., B.A. (Hons-JMC)
19.	Madrasa Qasimia Islamia, Kachehri Road, Gaya	Alimiat	B.A., B.A. (Hons-JMC)
20.	Jamia Muzaharul Uloom, Patna	Fazilat	B.A., B.A. (Hons-JMC)
21.	Al mahadul Aali Lit Tadreeb fil Qaza wal ifta, Patna	Takhassus	M.A (Islamic Studies, Arabic, Urdu, Persian)

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web elements in the second row of the periodic table place their electrons in the 2n shell as well as the 1n shell for instance lithium li has three electrons two fill the 1 s orbital and the third is placed in the 2 s orbital giving an electron configuration of 1 s 2 2 s 1

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web predictions from reliable sources have been used for these elements grayed out electron numbers indicate subshells filled to their maximum bracketed noble gas symbols on the left represent inner configurations that are the same in each period written out these are he 2 helium 1s 2 ne 10 neon 1s 2 2s 2 2p 6

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