

1. What explains the very high melting and boiling point of water

- |                                                                            |                                                                              |
|----------------------------------------------------------------------------|------------------------------------------------------------------------------|
| <input type="radio"/> A Strong dipole-dipole bonds between water molecules | <input type="radio"/> B Dispersion forces which are present in all molecules |
| <input type="radio"/> C Strong hydrogen bonds between water molecules      | <input type="radio"/> D Asymmetrical shape of the polar bonds.               |

2. Does HCl have hydrogen bonding?

- |                            |                             |
|----------------------------|-----------------------------|
| <input type="radio"/> A no | <input type="radio"/> B yes |
|----------------------------|-----------------------------|

3. Does CH<sub>4</sub> have hydrogen bonding?

- |                             |                            |
|-----------------------------|----------------------------|
| <input type="radio"/> A yes | <input type="radio"/> B no |
|-----------------------------|----------------------------|

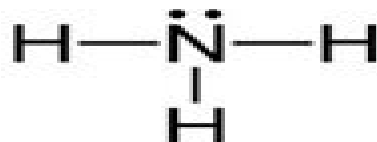
4. All molecules have London forces between them, but dipole-dipole and hydrogen bonding are so much stronger that when they are present we can ignore London forces. Which of these has ONLY London forces?

- |                                         |                                          |
|-----------------------------------------|------------------------------------------|
| <input type="radio"/> A I <sub>2</sub>  | <input type="radio"/> B OCl <sub>2</sub> |
| <input type="radio"/> C NH <sub>3</sub> | <input type="radio"/> D SH <sub>2</sub>  |

5. Which substance would have the weakest intermolecular forces of attraction?

- |                                         |                                          |
|-----------------------------------------|------------------------------------------|
| <input type="radio"/> A CH <sub>4</sub> | <input type="radio"/> B H <sub>2</sub> O |
| <input type="radio"/> C NaCl            | <input type="radio"/> D MgF <sub>2</sub> |

6. Intermolecular forces for: NH<sub>3</sub>



- |                                       |                                          |
|---------------------------------------|------------------------------------------|
| <input type="radio"/> A Dipole dipole | <input type="radio"/> B Dispersion Force |
|---------------------------------------|------------------------------------------|

# States Of Matter And Intermolecular Forces Test

**S.K. Jain & Shailesh K. Jain**



## States Of Matter And Intermolecular Forces Test:

STATES OF MATTER NARAYAN CHANGDER,2024-05-02 Note Anyone can request the PDF version of this practice set workbook by emailing me at cbsenet4u gmail com You can also get full PDF books in quiz format on our youtube channel <https://www.youtube.com/smartquiziz> I will send you a PDF version of this workbook This book has been designed for candidates preparing for various competitive examinations It contains many objective questions specifically designed for different exams Answer keys are provided at the end of each page It will undoubtedly serve as the best preparation material for aspirants This book is an engaging quiz eBook for all and offers something for everyone This book will satisfy the curiosity of most students while also challenging their trivia skills and introducing them to new information Use this invaluable book to test your subject matter expertise Multiple choice exams are a common assessment method that all prospective candidates must be familiar with in today's academic environment Although the majority of students are accustomed to this MCQ format many are not well versed in it To achieve success in MCQ tests quizzes and trivia challenges one requires test taking techniques and skills in addition to subject knowledge It also provides you with the skills and information you need to achieve a good score in challenging tests or competitive examinations Whether you have studied the subject on your own read for pleasure or completed coursework it will assess your knowledge and prepare you for competitive exams quizzes trivia and more

*States Of Matter* A. Goel,2006 This book has been written for the students of under graduate and post graduate level of the various universities in India A special feature of the book is that the text has been illustrated with a large number of line diagrams and the data presented in the form of numerous tables for reference and comparison In the preparation of text standard works and review by renowned author have been freely consulted and the reference given chapter wise At the end of the book will be found useful by those who wish to make a more detailed study of the topics discussed Contents Liquid State Solids State Gaseous State

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**Conceptual Chemistry Volume I For Class XI** S.K. Jain & Shailesh K. Jain,1998 Conceptual Chemistry Volume I For Class XI

(Chemistry) Physical Chemistry: States of Matter and Ionic Equilibrium Dr. MD. Faiyaz Ahsan,Dr. Nilima,2024-01-01 Buy Latest Chemistry Physical Chemistry States of Matter and Ionic Equilibrium e Book in English Edition for B Sc 2nd Semester Bihar State By Thakur publication

**Chemistry Class 11** Dr. S C Rastogi, ,Er. Meera Goyal,2022-09-29 Syllabus Unit I Some Basic Concepts of Chemistry Unit II Structure of Atom Unit III Classification of Elements and Periodicity in Properties Unit IV Chemical Bonding and Molecular Structure Unit V States of Matter Gases and Liquids Unit VI Chemical Thermodynamics Unit VII Equilibrium Unit VIII Redox Reactions Unit IX Hydrogen Unit X s Block Elements Alkali and Alkaline earth metals Group 1 and Group 2 Elements Unit XI Some p Block Elements General Introduction to p Block Elements Unit XII Organic Chemistry Some Basic Principles and Techniques Unit XIII Hydrocarbons

Classification of Hydrocarbons Unit XI V Environmental Chemistry Content 1 Some Basic Concepts of Chemistry 2 Structure of Atom 3 Classification of Elements and Periodicity in Properties 4 Chemical Bonding and Molecular Structure 5 States of Matter 6 Thermodynamics 7 Equilibrium 8 Redox Reactions 9 Hydrogen 10 s Block Elements 11 p Block Elements 12 Organic Chemistry Some Basic Principles and Techniques 13 Hydrocarbons 14 Environmental Chemistry I Appendix II Log antilog Table

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Sevian, 2013-07-09 Bringing together a wide collection of ideas reviews analyses and new research on particulate and structural concepts of matter Concepts of Matter in Science Education informs practice from pre school through graduate school learning and teaching and aims to inspire progress in science education The expert contributors offer a range of reviews and critical analyses of related literature and in depth analysis of specific issues as well as new research Among the themes covered are learning progressions for teaching a particle model of matter the mental models of both students and

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## **States Of Matter And Intermolecular Forces Test** Book Review: Unveiling the Power of Words

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quiz course 1 1k views definitions making a scaled version of a solid creates a similar solid the scaling happens by multiplying each defining dimension by the same number this number is

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in geometry a similar solid is an important concept students need to have a clear understanding of this concept as it may help them in their daily lives so what are similar solids these are two or more shapes that are similar in shape but their sizes are not the same

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we can identify similar solids by observing their corresponding dimensions if the corresponding dimensions are proportional then the solids are said to be similar follow the given steps to identify similar solids step 1 determine the dimensions that correspond step 2 calculate the length ratio of the corresponding dimensions

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similar figures two surfaces or solids are similar if any two corresponding sides or planes are proportional in similar figures of any kind pairs of corresponding line segments such as  $x_1$   $x_2$  and  $y_1$   $y_2$  have the same ratio  $x_1/x_2 = y_1/y_2$

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the diagram shows two similar solids and solid a has a volume of 80 cm<sup>3</sup> work out the volume of solid b

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calculate the volume of the larger cube using the illustration of the two similar solids indicated in the following illustration  
answers eq text 4 860 yd 3 eq

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two solids are similar if and only if they are the same type of solid and their corresponding linear measures radii heights base lengths etc are proportional surface area recall that when two shapes are similar the ratio of the area is the square of the scale factor

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created by ddufresne16 terms in this set 6 two solids are similar if and only if their bases are similar and their corresponding lengths are proportional if the scale factor of two similar solids is a b then the ratio of corresponding perimeters is a b we have an expert written solution to this problem

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steps to identify similar solids step 1 determine the corresponding dimensions step 2 find the ratio of lengths of the corresponding dimensions step 3 check if the all of the ratios are all

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play this game to review geometry are the two figures similar if yes what is the similarity ratio

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kuta software infinite geometry name similar solids date period are the two figures similar if so state the scale factor 1 24 ft 18 ft 12 ft 80 ft 40 ft 2 50 m 40 m some information about the surface area and volume of two similar solids has been given find the

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similar solid comparison worksheets how to compare similar solids the solid is anything having volume and surface area you may have the idea of similar solids by now we can compare all such objects if we are able to find their volume and surface area here is a way to discover how shapes are similar to each other

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jan 21 2020 determine whether two solids are similar by finding scale factors if possible use the similar solids theorem to find the surface area and volume of similar solids use a scale factor of a similar solid to find the missing side lengths video lesson examples 58 min introduction 00 00 28 determine if the solids are similar

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