1 (40	La Chatelier's Principle:			
		If a system in equitibrium is subjected to a constraint, a process will occur (1) which tends to countered the constraint; (AW)		
((0))	(10	While lower temperatures then 450°C would give a high yield SO <sub>2</sub> , they would make the reaction rate too show; (AW)	(3)	
	3000	Cooling is necessary to favour the forward <u>existential</u> reaction/AW	CUE	
	(9810)	Cost of chicker pipes to withouted the higher presumes will be too higher AW	$i \sim 10$	
(4)	60	King ton	(1)	
110,0	è	with a semi-forter at a man 1/en + for an	(1)	
	(00)	(I) No change in K <sub>p</sub> value:	(10)	
		Increase in partial pressure of SO, is counter-balanced by incitis partial pressure of SO,/	WHAT.	
		Extra SO <sub>3</sub> reacts to produce more SO <sub>3</sub> used equilibrium in re-established AW:	(10	
		(11)		
		Addition of V <sub>2</sub> O <sub>4</sub> catalyst has no effect on K <sub>2</sub> 1.	200	
		Catalyst only increases rate of straining equilibrium, without changing proportions of reactains and products/AW.	(1)	
		citto		
		$K_a$ value decreases; Endothermic decomposition of SO, to form more SO <sub>2</sub> and C is favoured, thus lowering $K_{\mu^2}$	(1) (1) (1) (Total: 12)	
	(6) (6)	(b) (b) (b) (c) (c) (c) (c) (c) (c)	If a system in equilibrium is subjected to a constraint, a process will on which tends to construe the constraint (AW)  (b)  (i) While lower temperatures than 450°C would give a high yield soo, they would make the reaction can too alone, (AW)  (ii) Cooling is necessary to favour the forward guaranteesis sention, AW  (iii) Cost of thicker paper to withouted the ingher parameters will be too high. AW  (iii) Cost of thicker paper to withouted the ingher parameters will be too high. AW  (iii) (i) (ii) (ii) (iii) (i	

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**L** Reisser

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