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web learning objectives by the end of this section you will be able to define half life define dating calculate age of old objects by radioactive dating unstable nuclei decay however some nuclides decay faster than others for example radium and polonium discovered by the curies decay faster than uranium

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disintegrates over a period of 90 years how many grams of cs 137 would remain a 1 25 g b 0 125 g c 0 00125 g d 12 5 g

correct answer b 0 125 g q2 selenium 83 has a half life of 25 0 minutes

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web each radioactive isotope will have its own unique half life that is independent of any of these factors figure pageindex 1

for cobalt 60 which has a half life of 5 27 years 50 remains after 5 27 years one half life 25 remains after 10 54 years two half

lives 12 5 remains after 15 81 years three half lives and so on

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remains after 3 0 years a 3 5 years b 4 5 y

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web half life problems name block 1 an isotope of cesium cesium 137 has a half life of 30 years if 1 0 g of cesium 137

disintegrates over a period of 90 years how many g of cesium 137 would remain a we r 2 actinium 226 has a half life of 29

hours

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web we utilize the equation that relate amount remaining initial mass and number of half lives $n = \frac{t}{T} \times n_0$ 1 2 4 x 50 3

125 g 5 what is the half life of an isotope that is 80 remained after 16 days solution remaining 80 100 therefore $n = \frac{t}{T} \times n_0$ 100

now using the half life equation in example 1 we have

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web what is the half life of this radioisotope after 2 days 5 grams of the radioisotope 1 day b 2 days c 5 days d 4 days 3 the

ratio of uranium 238 to lead 206 in a mineral is used to determine age b density c solubility d composition 4 a radioactive

isotope has a half life of 10 years

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web chapter 15 nuclear chemistry half life learning objectives define half life determine the amount of radioactive substance remaining after a given number of half lives whether or not a given isotope is radioactive is a characteristic of that particular isotope

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web section learning objectives by the end of this section you will be able to do the following explain radioactive half life and its role in radiometric dating calculate radioactive half life and solve problems associated with radiometric dating section key terms half life and the rate of radioactive decay unstable nuclei decay

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web solution if we compare the time that has passed to the isotope s half life we note that 44 0 s is exactly 4 half lives so using the previous equation $n = 4$ substituting and solving results in the following less than one third of a gram of ^{20}F remains skill building exercise the half life of ^{44}Ti is 60 0 y

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web solution 7 2 2 4 3 half lives 1 2 0 125 the amount remaining after 3 half lives $100.0\text{ g} \times 0.125 = 12.5\text{ g}$ remaining problem 2 Pd-100 has a half life of 3 6 days if one had 6.02×10^{23} atoms at the start how many atoms would be present after 20 0 days solution $20.0 \times 0.365 = 7.3\text{ days}$ half lives

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