Modeled after AP 2013 921:

2 C/Majot + 7 Outot ++ 4 COviet + 6 M/OSS

In the reaction represented above, what is the total number of moles of reactants consumed when 2.00 mole of COulg) is produced?

- At 45 SQ result
- 1.50 model
- C. 4.50 mod
- D. Schillered
- Cometed + 6 Outpot -> 4 COutpot + 4 Hu/CRT

In the reaction represented above, what is the total number of moles of reactants consumed when 2.00 recie of CO-dol is produced?

- A. D.Schmail
- B. 1.00 mod
- 3.50 molt
- D. T-DO-House
- 2 C/Hubbl + 7 Oubbl + 4 COUbbl + 6 HyOR5

In the reaction represented above, what is the total number of moles of products made when 3.50 mole of 0-lot is operamed?

- A 2 D3 mod
- the Contractor of
- C. 5.00 mod
- D. 20.0 west
- Cultividge + 6-Outge) -> 4 COutge + 8 HyDUD.

In the reaction represented above, what is the total number of moles of products made when 1.50 mole of O./gl is consumed?

- A. S.DOresell
- The State of State of
- 4.00 ened
- Dr. BullOtterwood

Mindeled after AP 2014a BUS

5. 4 NH date + 5 Culati -+ 4 NOBEL + 6 HuCOS

A 7.0 mol sample of Ningg) and a 90, mol sample of Oug) are placed in a previously evacuated container and react according to the equation above. How many makes of NOtg) are produced when one of the reactants has been totally consumed?

- A. J.S. mod.
- B. T.D. Hope
- C. Bull mode
- The Talk revisit
- Camplet = 3 Holet → 2 Chillet

Samples containing 4.0 mol Cytuigt and 5.0 mol Huigt are placed in a previously evacuated container and allowed to react according to the equation above. When one of the reactants is completely consumed, how much product is formed?

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Mina Lebitz

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