

# Solution Stoichiometry

1. What volume of a 0.35M  $\text{AgNO}_3$  is required to completely react with 65 ml of a 0.29M  $\text{NaCl}$  solution?



Required: V of  $\text{AgNO}_3$

Start with the other substance

$$\frac{0.29 \text{ mol NaCl}}{1} \cdot \frac{0.055 \text{ L NaCl}}{1} \cdot \frac{1 \text{ mol AgNO}_3}{1 \text{ mol NaCl}}$$

then convert

mol  $\rightarrow$  L

$$\cdot \frac{1 \text{ L AgNO}_3}{0.35 \text{ mol AgNO}_3}$$

$$= 0.0377 \text{ L} = 37.7 \text{ mL}$$

2. 37 mL of a 0.29M  $\text{KI}$  solution is required to completely react with 42 mL of a  $\text{Pb}(\text{NO}_3)_2$  solution. What was the original concentration of the lead (II) Nitrate solution?



Given:

$V_2$  37 mL  $\text{KI}$  soln  
 $M_2$  0.29M  $\text{KI}$  soln  
 $M_1$  42 mL  $\text{Pb}(\text{NO}_3)_2$

Required:

original concentration of the lead

$$\frac{0.29 \text{ mol KI}}{1} \cdot \frac{0.037 \text{ L}}{1} \cdot \frac{1 \text{ mol Pb}(\text{NO}_3)_2}{2 \text{ mol KI}} \cdot \frac{1}{0.042 \text{ L}}$$

$$= 0.1277$$

$$2(42) M_2 = 1(0.29)(37)$$

$$M_2 = 10.75 / 84$$

$$M_2 = 0.1277 \text{ M}$$

3. What mass of solid Magnesium Hydroxide can be produced if 15 mL of a 0.43M  $\text{Mg}(\text{NO}_3)_2$  solution reacts completely with excess  $\text{NaOH}$ ?



Required: mass of solid Magnesium Hydroxide if 15 mL of a 0.43M  $\text{Mg}(\text{NO}_3)_2$

Given: 15 mL  $\text{Mg}(\text{NO}_3)_2$ , 0.43M  $\text{Mg}(\text{NO}_3)_2$

$$\frac{0.43 \text{ mol Mg}(\text{NO}_3)_2}{1} \cdot \frac{0.015 \text{ L}}{1} \cdot \frac{1 \text{ mol Mg}(\text{OH})_2}{1 \text{ mol Mg}(\text{NO}_3)_2}$$

$$\cdot \frac{58.32 \text{ g}}{1 \text{ mol Mg}(\text{OH})_2} = 1.65 \text{ g Mg}(\text{OH})_2$$

4. What mass of solid Calcium Phosphate can be produced if 78 mL of a 0.17M Sodium Phosphate solution is mixed with 58 mL of a 0.29M Calcium chloride solution?



Given:

78 mL  $\text{Na}_3\text{PO}_4$   
0.17M  $\text{Na}_3\text{PO}_4$

58 mL  $\text{CaCl}_2$   
0.29M  $\text{CaCl}_2$

Required:

mass (g)  $\text{Ca}_3(\text{PO}_4)_2$

$$\frac{0.17 \text{ mol Na}_3\text{PO}_4}{1} \cdot \frac{0.078 \text{ L}}{1} \cdot \frac{1 \text{ mol Ca}_3(\text{PO}_4)_2}{2 \text{ mol Na}_3\text{PO}_4}$$

$$\cdot \frac{310.18 \text{ g Ca}_3(\text{PO}_4)_2}{1 \text{ mol Ca}_3(\text{PO}_4)_2} = 1.925 \text{ g Ca}_3\text{PO}_4$$

From  $\text{CaCl}_2$

$$\frac{0.29 \text{ mol CaCl}_2}{1} \cdot \frac{0.058 \text{ L}}{1} \cdot \frac{1 \text{ mol Ca}_3(\text{PO}_4)_2}{3 \text{ mol CaCl}_2} \cdot \frac{310 \text{ g Ca}_3(\text{PO}_4)_2}{1 \text{ mol}}$$

$$= 1.9377 \text{ g Ca}_3(\text{PO}_4)_2$$

molar mass:

$$\begin{aligned} \text{Ca} &- 2 \cdot 40.08 \\ \text{P} &- 2 \cdot 30.97 \\ \text{O} &- 8 \cdot 16 \end{aligned} \quad \left. \begin{array}{l} \\ \\ \end{array} \right\} 310.18 \text{ g/mol}$$

# Solution Stoichiometry Practice Problems

**Terry C. Jones**



## **Solution Stoichiometry Practice Problems:**

Survival Guide to General Chemistry Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics. The focus is not to recap or review the theoretical concepts well described in the available texts. Instead, the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry. Explanations, instructional process sequences, solved examples, and completely solved practice problems are greatly expanded, containing significantly more detail than can usually be devoted to in a comprehensive text. Many chapters also provide alternative viewpoints as an aid to understanding. Key Features: The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester. Each is written in a specific and detailed step by step process for problem solving, whether mathematical or conceptual. Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts. Includes a chapter designed to eliminate confusion concerning acid base reactions, which often persists through working with acid base equilibrium. Many chapters provide alternative viewpoints as an aid to understanding. This book addresses a very real need for a large number of incoming freshmen in STEM fields. Jacaranda Chemistry 1 VCE Units 1 and 2, learnON and Print Neale Taylor, Angela Stubbs, Robert Stokes, 2022-11-30 Developed by expert Victorian teachers for VCE students. The NEW Jacaranda Chemistry VCE series continues to deliver curriculum aligned material that caters to students of all abilities. Our expert author team of practising teachers and assessors ensures 100% coverage of the new VCE Chemistry Study Design 2023/2027. **The Practice of Chemistry Study Guide & Solutions Manual** Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes. Also includes solutions and explanations to odd numbered exercises. **Chemistry: 1,001 Practice Problems For Dummies (+ Free Online Practice)** Heather Hattori, Richard H. Langley, 2014-03-11 Practice makes perfect and helps deepen your understanding of chemistry. Every high school requires a course in chemistry, and many universities require the course for majors in medicine, engineering, biology, and various other sciences. 1001 Chemistry Practice Problems For Dummies provides students of this popular course the chance to practice what they learn in class, deepening their understanding of the material and allowing for supplemental explanation of difficult topics. 1001 Chemistry Practice Problems For Dummies takes you beyond the instruction and guidance offered in Chemistry For Dummies, giving you 1,001 opportunities to practice solving problems from the major topics in chemistry. Plus, an online component provides you with a collection of chemistry problems presented in multiple choice format to further help you test your skills as you go. Gives you a chance to practice and reinforce the skills you learn in chemistry class. Helps you refine your understanding of chemistry. Practice problems with answer explanations that detail every step of every problem. Whether you're studying chemistry at the high school, college, or graduate level, the practice problems in 1001

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**CliffsNotes Chemistry Practice Pack** Charles Henrickson, 2010-02-08 About the Contents Pretest Helps you pinpoint where you need the most help Topic Area Reviews Measurement and Units of Measurement Matter Elements Compounds and Mixtures Atoms I The Basics Formulas and Names of Ionic Compounds Acids and Bases The Mole Elements and Compounds Percent Composition and Empirical and Molecular Formulas Chemical Reactions and Chemical Equations Calculations Using Balanced Equations Atoms II Atomic Structure and Periodic Properties Chemical Bonding The Formation of Compounds Gases and the Gas Laws The Forces between Molecules Solids and Liquids Solutions and Solution Composition Acids Bases and Neutralization Glossary Customized Full Length Exam Covers all subject areas Pretest that pinpoints what you need to study most Clear concise reviews of every topic Targeted example problems in every chapter with solutions and explanations Customized full length exam that adapts to your skill level

[Multiple Solution Methods for Teaching Science in the Classroom](#) Stephen DeMeo, 2008 For the first time in science education the subject of multiple solution methods is explored in book form While a multiple method teaching approach is utilized extensively in math education there are very few journal articles and no texts written on this topic in science Teaching multiple methods to science students in order to solve quantitative word problems is

important for two reasons First it challenges the practice by teachers that one specific method should be used when solving problems Secondly it calls into question the belief that multiple methods would confuse students and retard their learning Using a case study approach and informed by research conducted by the author this book claims that providing students with a choice of methods as well as requiring additional methods as a way to validate results can be beneficial to student learning A close reading of the literature reveals that time spent on elucidating concepts rather than on algorithmic methodologies is a critical issue when trying to have students solve problems with understanding It is argued that conceptual understanding can be enhanced through the use of multiple methods in an environment where students can compare evaluate and verbally discuss competing methodologies through the facilitation of the instructor This book focuses on two very useful methods proportional reasoning PR and dimensional analysis DA These two methods are important because they can be used to solve a large number of problems in all of the four academic sciences biology chemistry physics and earth science This book concludes with a plan to integrate DA and PR into the academic science curriculum starting in late elementary school through to the introductory college level A challenge is presented to teachers as well as to textbook writers who rely on the single method paradigm to consider an alternative way to teach scientific problem solving

*Chemistry for Nerds Guide Book: Chemistry, Science, Nerd, Geek, Textbook, Guidebook, Study Guide, Educational, STEM, Science Gift* Matt Kingsley, Calling all curious minds and science enthusiasts Are you fascinated by the invisible forces that shape our world Do you crave a deeper understanding of the elements molecules and reactions that make up everything around us Then look no further than Chemistry for Nerds Unleash Your Inner Mad Scientist This isn't your typical boring textbook This is a thrilling adventure through the captivating world of chemistry written in a way that's engaging accessible and downright fun Inside these pages you'll discover The secrets of matter From atoms and molecules to the states of matter and the laws that govern them The magic of reactions Explore the explosive world of chemical reactions from kinetics and equilibrium to acids bases and buffers The wonders of the elements Unravel the mysteries of the periodic table and the trends that connect its diverse inhabitants The power of chemistry in action See how chemistry shapes our environment fuels our technologies and even sustains life itself Chemistry for Nerds is packed with Crystal clear explanations Complex concepts are broken down into bite sized pieces making even the most challenging topics easy to grasp Engaging examples and analogies Relate chemistry to everyday life with fun and memorable examples Expert practical tips Put your knowledge into action with helpful tips and tricks for mastering chemistry concepts Whether you're a student a hobbyist or simply curious about the world around you Chemistry for Nerds will ignite your passion for science and unleash your inner mad scientist Get your copy today and start exploring the amazing world of chemistry

**Chemistry** Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas

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 John McMurry, 1998      *An Introductory Guide to EC Competition Law and Practice* Valentine Korah, 1994      **Chemistry**  
**Workbook For Dummies** Peter J. Mikulecky, Chris Hren, Christopher R. Hren, 2014-11-24 Hundreds of practice problems to help you conquer chemistry. Are you confounded by chemistry? Subject by subject problem by problem. Chemistry Workbook For Dummies lends a helping hand so you can make sense of this often intimidating subject. Packed with hundreds of practice problems that cover the gamut of everything you'll encounter in your introductory chemistry course, this hands-on guide will have you working your way through basic chemistry in no time. You can pick and choose the chapters and types of problems that challenge you the most or you can work from cover to cover. With plenty of practice problems on everything from matter and molecules to moles and measurements, Chemistry Workbook For Dummies has everything you need to score higher in chemistry. Practice on hundreds of beginning to advanced chemistry problems. Review key chemistry concepts. Get complete answer explanations for all problems. Focus on the exact topics of a typical introductory chemistry course. If you're a chemistry student who gets lost halfway through a problem or worse yet doesn't know where to begin, Chemistry Workbook For Dummies is packed with chemistry practice problems that will have you conquering chemistry in a flash.

Basic Concepts of Chemistry Leo J. Malone, Theodore Dolter, 2008-12-03 Engineers who need to have a better understanding of chemistry will benefit from this accessible book. It places a stronger emphasis on outcomes assessment which is the driving force for many of the new features. Each section focuses on the development and assessment of one or two specific objectives. Within each section a specific objective is included, an anticipatory set to orient the reader, content discussion from established authors and guided practice problems for relevant objectives. These features are followed by a set of independent practice problems. The expanded Making it Real feature showcases topics of current interest relating to the subject at hand such as chemical forensics and more medical related topics. Numerous worked examples in the text now include Analysis and Synthesis sections which allow engineers to explore concepts in greater depth and discuss outside relevance.

*Foundations of College Chemistry* Morris Hein, Susan Arena, 2013-01-01 Learning the fundamentals of chemistry can be a difficult task to undertake for health professionals. For over 35 years Foundations of College Chemistry Alternate 14th Edition has helped

readers master the chemistry skills they need to succeed It provides them with clear and logical explanations of chemical concepts and problem solving They ll learn how to apply concepts with the help of worked out examples In addition Chemistry in Action features and conceptual questions checks brings together the understanding of chemistry and relates chemistry to things health professionals experience on a regular basis

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Introductory Chemistry Charles H. Corwin, 2011 To help students learn chemical skills and concepts more effectively Introductory Chemistry Concepts and Critical Thinking Sixth Edition highlights the connection between key concepts and key problem solving skills through critical thinking Math and problem solving are covered early in the text Corwin builds your problem solving ability through innovative learning aids and technology formulated to meet your needs This revision retains all the strengths of the previous editions while adding emphasis on conceptual understanding and critical thinking

*A Visual Analogy Guide to Chemistry, 2e* Paul A Krieger, 2018-02-01 A Visual Analogy Guide to Chemistry is the latest in the innovative and widely used series of books by Paul Krieger This study guide delivers a big picture view of difficult concepts and effective study tools to help students learn and understand the details of general organic and biochemistry topics A Visual Analogy Guide to Chemistry is a worthwhile investment for any introductory chemistry student

General Chemistry, Reactions First Kevin Revell, 2024-12-04 Revell s General Chemistry empowers students to grasp essential topics and concepts with more ease Using a friendly approach the text uses metaphors and relatable examples to demystify even the most challenging subjects in general chemistry

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