

# Thermochemistry Study Guide

**Qualification:** Write the Scientific Notation & use it to answer the questions below

Scientific Notation

1. 100 grams of a substance at 25°C, calculate the specific heat

2. 10g of mercury at 25°C, calculate the specific heat

**Reading Science:** Label the graph

(Amount of heat)

Reaction



1. How much energy is released?

2. Calculate the energy released

**Calculations:** Answer the questions

1. 10g of water at 25°C, calculate the specific heat

2. 10g of mercury at 25°C, calculate the specific heat

Name \_\_\_\_\_ Period \_\_\_\_\_ Date \_\_\_\_\_

## Thermochemistry Study Guide

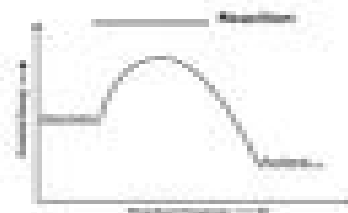
Fill in the blanks with the correct word from the following word bank to complete the statements.  
(Amount Heat, Heat, Chemical Equilibrium, Light, Thermodynamics, Energy, Temperature, Transformation, Thermal)

1. According to the law of \_\_\_\_\_, energy cannot be created or destroyed, only \_\_\_\_\_ from one form to another.
2. \_\_\_\_\_ is the ability to do work. \_\_\_\_\_ energy is stored in chemical bonds. \_\_\_\_\_ energy is due to the motion of particles and is felt as heat. \_\_\_\_\_ energy is one we can use and is created by electromagnetic waves.
3. \_\_\_\_\_ is the measurement of heat energy which can be measured using thermometers while \_\_\_\_\_ is the energy transferred due to temperature difference.
4. There is thermal \_\_\_\_\_ when both the system and the surroundings have the same temperature.

Identify which Potential Energy Diagram is Endothermic and which is Exothermic and fill in the blanks to compare their properties.



Heat is \_\_\_\_\_  
\_\_\_\_\_ Chemical Bonds  
Q is \_\_\_\_\_  
Example: \_\_\_\_\_



Heat is \_\_\_\_\_  
\_\_\_\_\_ Chemical Bonds  
Q is \_\_\_\_\_  
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# Solutions And Thermochemistry Study Guide

**F Rizvi**



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used for the same properties This unfortunate situation is particularly obvious in those chapters where partial specific or molar quantities had to be introduced however it also occurs in those contributions concerning phase changes of macromolecules      *Summer School Number* Kansas State College, 1920      **The Athenaeum** James Silk Buckingham, John Sterling, Frederick Denison Maurice, Henry Stebbing, Charles Wentworth Dilke, Thomas Kibble Hervey, William Hepworth Dixon, Norman Maccoll, Vernon Horace Rendall, John Middleton Murry, 1899

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