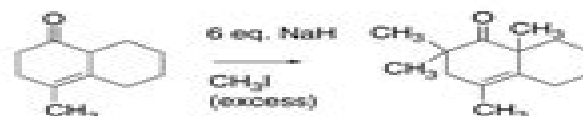


- b. This transformation involves methylation at all enolizable positions. The alkylation was effected using a sixfold excess of NaH and excess methyl iodide. Evidently there is not a significant amount of methylation at C(4), which could occur through γ -alkylation of the C(8a)-enolate.



- c. This alkylation was accomplished using two equivalents of NaNH₂ in liquid NH₃. The more basic site in the dianion is selectively alkylated. Note that the dianion is an indenyl anion, and this may contribute to its accessibility by di-deprotonation.



- d. This is a nitrile alkylation involving an anion that is somewhat stabilized by conjugation with the indole ring. The anion was formed using NaNH₂ in liquid NH₃.



1. The reaction was carried out using 100% NaH and 100% MeI in 100% Et₂O. The reaction was carried out at 0°C for 24 hours. The reaction was carried out in a 100 mL round-bottom flask equipped with a magnetic stirrer. The reaction mixture was poured into 100 mL of water and extracted with 100 mL of Et₂O. The organic layer was dried over anhydrous Na₂SO₄ and concentrated under reduced pressure. The residue was purified by column chromatography on silica gel using 10% Et₂O in hexanes as the eluent. The pure product was obtained as a colorless oil.



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Lingjun Ying



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