

6. Write the equilibrium expression in terms of partial pressures (K_p) for each of the following reactions.

Rate the reactions in order of their increasing tendency to proceed toward completion:

b c d a



$K_p = 1 \times 10^{28} \text{ atm}$

$$K_p = \frac{(P_{\text{N}_2})^2 (P_{\text{H}_2\text{O}})^6}{(P_{\text{NH}_3})^4 (P_{\text{O}_2})^3}$$



$K_p = 5 \times 10^{-11}$

$$K_p = \frac{(P_{\text{NO}})^2}{(P_{\text{N}_2})(P_{\text{O}_2})}$$



$K_p = 1 \times 10^{-13}$

$$K_p = \frac{(P_{\text{H}_2})(P_{\text{F}_2})}{(P_{\text{HF}})^2}$$



$K_p = 4.7 \times 10^{-4} \text{ atm}$

$$K_p = \frac{(P_{\text{NO}})^2 (P_{\text{Cl}_2})}{(P_{\text{NOCl}})^2}$$

A Question That You Should Be Able To Answer:
Why don't the K_p 's in (b) and (c) have units?

(b) $\frac{\text{atm}^2}{\text{atm}^2} = 1$ (c) $\frac{\text{atm}^2}{\text{atm}^2} = 1$

7. (a) Write the K_c expression for
 $2\text{SO}_2(\text{g}) + \text{O}_2(\text{g}) \rightleftharpoons 2\text{SO}_3(\text{g})$

$$K_c = \frac{[\text{SO}_3]^2}{[\text{SO}_2]^2 [\text{O}_2]}$$

Calculate the value of K_c :

At equilibrium: $[\text{SO}_2] = 1.50 \text{ M}$
 $[\text{O}_2] = 1.25 \text{ M}$
 $[\text{SO}_3] = 3.50 \text{ M}$

$$= \frac{(3.50)^2}{(1.50)^2 (1.25)} = 4.355$$

$$= 4.36 \text{ M}^{-1}$$

- (b) If we reverse the equation, it is:



Write the K_c expression for this equation and calculate the new value of K_c :

$$K_c' = \frac{[\text{SO}_2]^2 [\text{O}_2]}{[\text{SO}_3]^2} = \frac{(1.5)^2 (1.25)}{(3.50)^2} = .2259$$

How does the expression and the value of K_c in 7(b) compare with those in 7(a)?

$$K_c' = \frac{1}{K_c}$$

- (c) If we now multiply all of the coefficients by $\frac{1}{2}$:



Write the K_c expression for this equation and calculate the new value of K_c :

$$K_c'' = \frac{[\text{SO}_2] [\text{O}_2]^{1/2}}{[\text{SO}_3]} = \frac{(1.5) (1.25)^{1/2}}{(3.50)} = .479$$

How do they compare with 7(b)?

$$K_c'' = \sqrt{K_c'}$$

- (d) What would happen to the K_c expression and its value if we doubled the coefficients?

$$K_c''' = (K_c')^2$$

Summarize:

Equation	K_c expression & Value
doubled	SQUARE
reversed	INVERSE
halved	SQ. ROOT

8. Consider an equilibrium that occurs in two steps:



- (a) Write the overall reaction.

(b) How do the K_c 's for the two steps (K_{c1} & K_{c2}) relate to the K_c of the overall reaction (K_c)?

$$K_{c1} \cdot K_{c2} = K_c$$

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RJ Shavelson



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 United States. Agricultural Research Service, 1969 **Millikan's School: A History of the California Institute of Technology** Judith Goodstein, 2020-10-19 In November 1891 wealthy former abolitionist and Chicago politician Amos Throop founded a thoroughly undistinguished small college in Pasadena California which he named after himself Millikan's School is the history of this institution that stands today at the pinnacle of world academics with 300 full time faculty nearly 1 000 undergraduate 1 250 graduate students and 39 Caltech and alumni Nobel Prize recipients Although Amos Throop the name of the college was changed to Caltech in 1920 could not have realized the importance of geography the fact that Pasadena lay at the foot of Mount Wilson was central to its success astronomer George Ellery Hale built his telescope there in 1902 the finest at that time in the world Later Hale joined the board of trustees of the struggling school and persuaded Arthur Amos Noyes former president of MIT and the nation's leading physical chemist to join him in Pasadena The third member of Caltech's founding troika was renowned physicist Robert A Millikan from the University of Chicago The dedication of Caltech in 1920 and the proclamation of what it stood for in science and education set the stage for Millikan who functioned as the school's president to bring the best and the brightest from all over the world Theodore von Kármán in aeronautics Thomas Hunt Morgan in biology Paul Sophus Epstein in physics Benoit Gutenberg in seismology Linus Pauling in chemistry to Pasadena to work in an ever larger number of areas in science and technology The book also covers the funding planning and construction of the 200 inch telescope on Palomar Mountain Willy Fowler's work in nuclear astrophysics and the wartime rocket experiments that grew into the Jet Propulsion Laboratory JPL today the world leader in deep space exploration Millikan's School presents an interesting and thoroughly reliable account of the astonishing change over a period of a few years of a small technical school in Pasadena California into one of the world's leading scientific institutions Linus Pauling In Millikan's School Judith Goodstein tells the remarkable story of the rise of Caltech She details how Millikan aided by Hale and Arthur Amos Noyes America's leading physical chemist and another of Hale's inspired acquisitions took a former trade school and forged from it a grandiose university among the orange groves It would be impossible while reading Goodstein's lively account not to be impressed by the energy drive and boundless enthusiasm of men like Millikan Hale and Noyes who had the bare faced audacity to set about building an institute to rival the cream of the universities of Europe and America Marcus Chown New Scientist Goodstein's story is first and foremost the tale of three men the astronomer George Ellery Hale the chemist Alfred Noyes and the physicist Robert Millikan It is the story of their attempts to transform an undistinguished little school founded in 1891 into a world class scientific establishment A useful book Tony Rothman Science In Millikan's School the story of Throop University's transformation into Caltech is told with precision Judith Goodstein's history offers a quick tour of the landmarks of science in the mid 20th Century and a glance at how pure science puts itself at the service of government commerce and the military Goodstein approaches her subject with a healthy sense of humor and an acute sense

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