-4NaOH	$+2F_{2}$	$\rightarrow 4 NaF$	+2H	$_{2}O + O_{2}$
ii) (conc)				

Reaction with Cl₂, Br₂ and I₂:

$$2NaOH + X_2 \rightarrow NaX + NaOX + H_2O$$

0 (cold / dilute)

ii)
$$6NaOH + 3X_2 \rightarrow 5NaX + NaXO_1 + 3H_2O$$

Sodium carbonate (Na₂CO₃.10H₂O) called washing soda.

Anhydrous sodium carbonate is called soda ash.

Sodium carbonate is prepared by

(i) Le-Blanc process (ii) Ammonia soda process or Solvay process (iii) Electrolytic process

- Le-Blanc process: The raw materials used are NaCl. cone. H₂SO₄, lime stone
- Na₂CO₃ + CaS is called as black ash.
- Solvays process: The raw materials used are NaCl. NH₃ and CaCO₃.

The by product in this process is $CaCl_2$.

- Sodium bi carbonate (NaHCO₃) is called as baking soda.
- Na₂CO₃ + H₂O + CO₂ → 2NaHCO₃
- Aqueous solution of NaHCO₃ is alkaline due to anionic hydrolysis (HCO₃).
- Aqueous solution of NaHCO₃ cannot give any colour with phenolphthalein.
- The oxides of Be, Mg, Ca, Sr and Ba are basic therefore these elements are called alkaline earth metals.
- Radium is radio active element.

- The outer electronic configuration is _{RS}². They exhibit +2 oxidation state.
- The most abundant alkaline earth metal is calcium.
- Be and Mg do not give any flame colouration.
- Be shows diagonal relationship with A1.
- Alkaline earth metals form binary compounds with hydrogen called hydrides.

The ionic nature of hydrides increases in the order

$$BeH_2 < MgH_2 < CaH_2 < SrH_2 < BaH_2$$
 (Ionic)

- Alkaline earth metals react with halogens to form their respective halides.
- Except Berylium halides, all the other halides of alkaline earth metals are ionic.
- Alkaline earth metal fluorides are generally insoluble in water except BeF₂.
- Mg is used in the manufacture of alloys:
 - (i) Magnalium: A1 + Mg (85 - 90%) (1 - 15%)
 - (ii) Elektron: Mg + Zn (95%) (5%)
- Gypsum on heating to 120-130⁰C gives plaster of paris. (2CaSO₄.H₂O).
- Plaster of paris sets to hard mass in two stages

$$CaSO_4\frac{1}{2}H_2O + \frac{3}{2}H_2O \xrightarrow{\text{sering inspir}} + CaSO_4 2H_2O$$

 $CaSO_4 \ 2H_2O$ hardware may $\to CaSO_4 \ 2H_2O$ (ortho rhombic di hydrate) (Monoclinic dihydrate)

- Lime mortar is a mixture of 1 part of slaked lime, 3 parts of sand and water.
- Mortar mixed with cement is called cement mortar.
- Lime stone and clay when heated together give hydraulic mortar. This is used for bleaching purposes and as an antiseptic.

World Of Chemistry 2 1 Answers

Don DeYoung, John Whitcomb, Donald B. DeYoung

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English Mechanic and World of Science ... ,1884 Norms of Word Association Leo Postman, Geoffrey Keppel, 2014-05-10 Norms of Word Association contains a heterogeneous collection of word association norms This book brings together nine sets of association norms that were collected independently at different times during a 15 year period Each chapter is a self contained unit The order in which the norms are presented is arbitrary although an attempt is made to group together norms that seem to belong together The 1952 Minnesota norms are presented first due to age and in recognition of the fact that a number of the norms that follow are direct outgrowths of this work. The next three norms in this collection are responses to the Russell Jenkins stimuli obtained from subjects representing different linguistic communities A summary of association norms collected from British and Australian subjects are reported along with association norms from German and French college students and French workmen Four sets of norms that are not directly related to the 1952 Minnesota collection are included The text will be of interest to historians and researchers in the field of verbal learning and The Journal of Education ,1893 The Position of the German Language in the World Ulrich verbal behavior Ammon, 2019-08-08 The Position of the German Language in the World focuses on the global position of German and the factors which work towards sustaining its use and utility for international communication From the perspective of the global language constellation the detailed data analysis of this substantial research project depicts German as an example of a second rank language The book also provides a model for analysis and description of international languages other than English It offers a framework for strengthening the position of languages such as Arabic Chinese French Portuguese Spanish and others and for countering exaggerated claims about the global monopoly position of English This comprehensive handbook of the state of the German language in the world was originally published in 2015 by Walter de Gruyter in German and has been critically acclaimed Suitable for scholars and researchers of the German language the handbook shows in detail how intricately and thoroughly German and other second rank languages are tied up with a great number of societies and how these statistics support or weaken the languages functions and maintenance Oxford EAP Pre-intermediate/B1 **Student Book** Edward de Chazal, John Hughes, 2015-08-13 Oxford English for Academic Purposes offers a specialist course covering listening speaking and reading in key areas of academic life such as lectures presentations and textbooks The course is consistent with levels A2 to C1 of the Common European Frame of Reference for the teaching of foreign languages Great downloadable resources to support you when using Oxford English for Academic Purposes can be found at https elt **Digest of Education Statistics 2006** Thomas D. Snyder, Sally A. Dillow, Charlene M. oup com student oxfordeap Hoffman, 2007

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