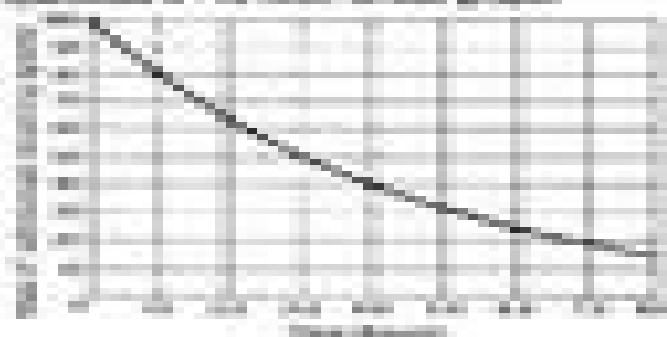


2-4- Nuclear Chemistry

AP EXAM OBJECTIVES

1. When Marguerite was born, her father (Mr. A.K.A. "Radioactive Father") was emitting alpha radiation.
(A) emitting the most "radioactivity"
(B) emitting alpha radiation
(C) emitting beta radiation
(D) emitting gamma radiation
2. When Marguerite's father was born, his father (Mr. A.K.A. "Radioactive Grandfather") was emitting alpha radiation.
(A) emitting the most "radioactivity"
(B) emitting alpha radiation
(C) emitting beta radiation
(D) emitting gamma radiation
3. Radioactive isotopes are "radioactive" because they emit particles or photons.
(A) Yes (B) No (C) Yes (D) No
(A) Yes (B) No (C) Yes (D) No
4. Which of the changes caused by alpha particles?
(A) $Z-2$ (B) $Z-1$ (C) $Z+1$ (D) $Z+2$
5. Which type of radiation is capable of being stopped by a sheet of paper?
(A) alpha (B) beta (C) gamma (D) neutron
6. What time does the sample spontaneously emit a half-life of 1.5 hours. What time of the following substance is most likely to be 1.5 hours?
(A) 1.5 (B) 3.0 (C) 4.5 (D) 6.0
(A) 1.5 (B) 3.0 (C) 4.5 (D) 6.0
(A) 1.5 (B) 3.0 (C) 4.5 (D) 6.0
(A) 1.5 (B) 3.0 (C) 4.5 (D) 6.0
7. In a decay series of the (U-238) the first three steps involve alpha decay. What is the result of these steps?
(A) $Z-2$ (B) $Z-1$ (C) $Z+1$ (D) $Z+2$
(A) $Z-2$ (B) $Z-1$ (C) $Z+1$ (D) $Z+2$

Questions 8 - 10 refer to the graph.



8. According to the above data, what is the half-life of the substance?
(A) 12.5 hours (B) 25 hours
(C) 37.5 hours (D) 50 hours
9. What percent of the original sample remains after 75 hours?
(A) 12.5% (B) 25%
(C) 37.5% (D) 50%
10. Which is the graph above, a curve for a substance with half-life of 2.5 hours.
(A) Yes (B) No (C) Yes (D) No
11. Radioactive U-238 has a half-life of 4.5 hours. What percent of a sample remains after 13.5 hours?
(A) 12.5% (B) 25% (C) 37.5% (D) 50%
12. Which of the following describes what occurs in the chemical process?
(A) is being oxidized or hydrogenated with hydrogen ions.
(B) is reduced to give ions a positive and an electron.
(C) ions are being oxidized and combined with a hydrogen ion.
(D) is positive and an electron is being combined with oxygen.

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Michael Seilmaier



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