

Titration Practice Worksheet

Find the requested quantities in the following problems:

- 1) If it takes 54 mL of 0.1 M NaOH to neutralize 125 mL of an HCl solution, what is the concentration of the HCl?

$$\begin{aligned} M_1 &= 0.1 \text{ M} \\ V_1 &= 54 \text{ mL} \\ M_2 &= ? \\ V_2 &= 125 \text{ mL} \end{aligned}$$

$$\begin{aligned} M_1 V_1 &= M_2 V_2 \\ (0.1 \text{ M})(54 \text{ mL}) &= (125 \text{ mL}) M_2 \\ M_2 &= \frac{(0.1 \text{ M})(54 \text{ mL})}{125 \text{ mL}} = 0.0432 \text{ M} = 4.32 \times 10^{-2} \text{ M} \end{aligned}$$

The concentration of HCl is $4.32 \times 10^{-2} \text{ M}$.

- 2) If it takes 25 mL of 0.05 M HCl to neutralize 345 mL of NaOH solution, what is the concentration of the NaOH solution?

$$\begin{aligned} M_1 &= 0.05 \text{ M} \\ V_1 &= 25 \text{ mL} \\ M_2 &= ? \\ V_2 &= 345 \text{ mL} \end{aligned}$$

$$\begin{aligned} M_1 V_1 &= M_2 V_2 \\ (0.05 \text{ M})(25 \text{ mL}) &= (345 \text{ mL}) M_2 \\ M_2 &= \frac{(0.05 \text{ M})(25 \text{ mL})}{345 \text{ mL}} = 0.00362 \text{ M} = 3.62 \times 10^{-3} \text{ M} \end{aligned}$$

The concentration of NaOH is $3.62 \times 10^{-3} \text{ M}$.

- 3) If it takes 50 mL of 0.5 M KOH solution to completely neutralize 125 mL of sulfuric acid solution (H_2SO_4), what is the concentration of the H_2SO_4 solution?



$$n = MV$$

$$n = (0.5 \text{ M})(50 \text{ mL}) = 25 \text{ mol OH}^-$$

2 times $[\text{OH}^-]$ is needed

to neutralize $[\text{H}^+]$. $\rightarrow [\text{H}^+] = \frac{1}{2} [\text{OH}^-]$

$$M = \frac{12.5 \text{ mol}}{125 \text{ mL}} = \boxed{0.10 \text{ M}}$$

- 4) Can I titrate a solution of unknown concentration and still get a meaningful answer? Explain your answer in a few sentences.

Titration cannot be done without molarity of at least one of the substances. In order to solve $M_1 V_1 = M_2 V_2$, both molarities cannot be unknown.

- 5) Explain the difference between an endpoint and equivalence point in a titration.

Endpoint: titration can be stopped.

Equivalence point: when $[\text{H}^+]$ and $[\text{OH}^-]$ are neutralized.

The effectiveness of a titration is measured by the closeness of the endpoint to the equivalence point.

Solution Practice Answer Key

Robert R. Redfield



Solution Practice Answer Key:

Solution Practice Answer Key: Bestsellers in 2023 The year 2023 has witnessed a noteworthy surge in literary brilliance, with numerous engrossing novels captivating the hearts of readers worldwide. Lets delve into the realm of popular books, exploring the fascinating narratives that have captivated audiences this year. Solution Practice Answer Key : Colleen Hoover "It Ends with Us" This poignant tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover skillfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can triumph. Uncover the Best : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This intriguing historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids absorbing storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Solution Practice Answer Key : Delia Owens "Where the Crawdads Sing" This captivating coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens spins a tale of resilience, survival, and the transformative power of nature, entrancing readers with its evocative prose and mesmerizing setting. These top-selling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of engaging stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a brilliant and gripping novel that will keep you guessing until the very end. The novel is a cautionary tale about the dangers of obsession and the power of evil.

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