PERIODIC TABLE TRENDS

Step 1: Complete the squares for each element by adding the atomic number, name, and atomic mass. Write the atomic number at the top of the square, . Be Write the atomic name under the symbol. Write the atomic mass at the bottom of the square. Step 2: Determine the number of electrons, protons, Bohr Diagram Dot Structure and neutrons in each statement. Be Step 3: Create a Bohr diagram for each element. -Metal Nonmetal Metalloid Step 4: Draw a Lewis Dot Structure for each element. Solid Cities.

Step 5: Show whether the element is a metal, nonmetal, or metalloid by circling the correct response. Also determine if the element is a solid or gas by circling the correct choice.

Step 6: Use the following colors to shade in the square for each element. You should ONLY color in the small square in the upper left-hand corner and not the entire card.

Green - Li and Na
Purple - F and Cl
Brown - N and P

Pink - O and S

Pink - O and S

Orange - B and Al
Yellow - He, Ne, and Ar

Blue - Be and Mg
Red - C and Si
White - H

Step 7: Cut the cards apart and arrange according to atomic number in the pattern shown below. Once you have the cards arranged in the correct order, glue them to a large sheet of construction paper.

 1
 2

 3
 4
 5
 6
 7
 8
 9
 10

 11
 12
 13
 14
 15
 16
 17
 18

Step 8: Put a title on your table.

Step 9: Make a key for each color with the name of the group for that column.

Step 10: Write the column number at the top of each group and the period number on the left side of the table. Be careful, column numbers are not in order from 1-18 and the transition metals, columns 3-12 are not on the table.

Step 11: Draw the red zigzag line that separates the metals from the nonmetals.

Step 12: Answer the questions on the worksheet using the information on your periodic table.

Starr Practice Periodic Table Answer Key

Tao Wei

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