

1. What explains the very high melting and boiling point of water

- | | | | |
|-------------------------|--|-------------------------|---|
| <input type="radio"/> A | Strong dipole-dipole bonds between water molecules | <input type="radio"/> B | Strong hydrogen bonds between water molecules |
| <input type="radio"/> C | Dispersion forces which are present in all molecules | <input type="radio"/> D | Asymmetrical shape of the polar bonds. |

2. Does CH₄ have hydrogen bonding?

- | | | | |
|-------------------------|----|-------------------------|-----|
| <input type="radio"/> A | no | <input type="radio"/> B | yes |
|-------------------------|----|-------------------------|-----|

3.  At higher temperatures

- | | | | |
|-------------------------|---|-------------------------|------------------------------------|
| <input type="radio"/> A | a gas contracts | <input type="radio"/> B | particles in an object move faster |
| <input type="radio"/> C | particles in an object have less energy | | |

4.  A gas

- | | | | |
|-------------------------|---|-------------------------|---------------------------|
| <input type="radio"/> A | has a definite shape but no definite volume | <input type="radio"/> B | has fast-moving particles |
| <input type="radio"/> C | has a definite volume but no definite shape | | |

5. Define a liquid.

- | | | | |
|-------------------------|-----------------------------------|-------------------------|----------------|
| <input type="radio"/> A | Definite volume, indefinite shape | <input type="radio"/> B | Found in stars |
|-------------------------|-----------------------------------|-------------------------|----------------|

States Of Matter And Intermolecular Forces Quiz

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States Of Matter And Intermolecular Forces Quiz:

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