

STATES OF MATTER

- All substances are made of matter. Matter is all the stuff around you – everything you can touch, see or smell. Anything that has **mass** (weight) and **volume** (occupies space) is matter. **Atom**:- is the smallest particle of matter. Everything is made up of atoms.
- The three states of matter are solid, liquid and gas.
- Solids, liquids and gases are all made of tiny particles. This is called the **particle theory**.
- The particles are attracted to each other. Some particles are attracted strongly to each other and others weakly.
- The particles move around. They are described as having kinetic energy. The kinetic energy of the particles increases with temperature.

Solids

Particles of solids are tightly packed, are very close together in a fixed arrangement.

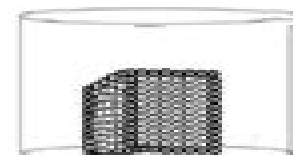
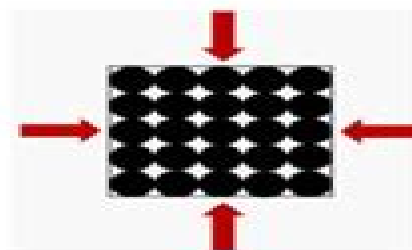
There are very strong forces of attraction between the particles holding them together.

There are very small spaces between the particles – high density.

This makes solids hard (usually) and they **cannot** be compressed.

Particles have a small amount of energy, particles vibrate but do not move around.

Solids have a definite shape and a definite volume.

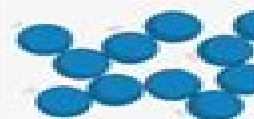


Liquids

Particles of liquids are tightly packed, but are far enough apart to slide over one another.

There are strong forces of attraction between the particles holding them together but not in an orderly arrangement.

There are small spaces between the particles – medium density.



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David Kirk



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1. Understanding the eBook States Of Matter Workbook Answers Pearson
 - The Rise of Digital Reading States Of Matter Workbook Answers Pearson
 - Advantages of eBooks Over Traditional Books
2. Identifying States Of Matter Workbook Answers Pearson
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an States Of Matter Workbook Answers Pearson
 - User-Friendly Interface
4. Exploring eBook Recommendations from States Of Matter Workbook Answers Pearson

- Personalized Recommendations
- States Of Matter Workbook Answers Pearson User Reviews and Ratings
- States Of Matter Workbook Answers Pearson and Bestseller Lists
- 5. Accessing States Of Matter Workbook Answers Pearson Free and Paid eBooks
 - States Of Matter Workbook Answers Pearson Public Domain eBooks
 - States Of Matter Workbook Answers Pearson eBook Subscription Services
 - States Of Matter Workbook Answers Pearson Budget-Friendly Options
- 6. Navigating States Of Matter Workbook Answers Pearson eBook Formats
 - ePub, PDF, MOBI, and More
 - States Of Matter Workbook Answers Pearson Compatibility with Devices
 - States Of Matter Workbook Answers Pearson Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of States Of Matter Workbook Answers Pearson
 - Highlighting and Note-Taking States Of Matter Workbook Answers Pearson
 - Interactive Elements States Of Matter Workbook Answers Pearson
- 8. Staying Engaged with States Of Matter Workbook Answers Pearson
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers States Of Matter Workbook Answers Pearson
- 9. Balancing eBooks and Physical Books States Of Matter Workbook Answers Pearson
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection States Of Matter Workbook Answers Pearson
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine States Of Matter Workbook Answers Pearson
 - Setting Reading Goals States Of Matter Workbook Answers Pearson
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of States Of Matter Workbook Answers Pearson

- Fact-Checking eBook Content of States Of Matter Workbook Answers Pearson
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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