

States of Matter & Phase Changes

Why?

Most substances go through a phase change when heated or cooled. Molecules of a substance are held together in either the solid, liquid, or gaseous phase by intermolecular forces. It is necessary to discuss what is occurring at the molecular level in order to explain how an ice cube is melted or how water is boiled.

New Concepts

- Heating / cooling curves
- Intermolecular Forces

Learning Objective:

- To determine what is occurring on the molecular level during a phase change

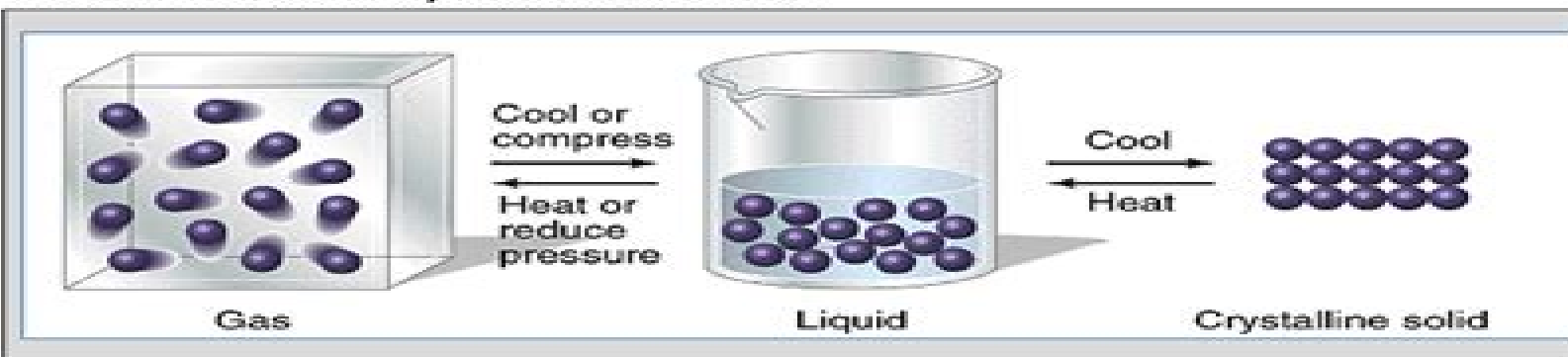
Success Criteria

- Students will be able to construct their own heating or cooling curve when given the temperatures at which phase changes occur.
- Students will be able to explain the heating curve on the micro-view.

Prerequisites

- Temperature (average kinetic energy)
- Phases of matter: solid, liquid, gas
- Molecules (particles)

Model 1: Kinetic Theory and States of Matter



1. What are the key characteristics of atoms and molecules in gases, liquids, and solids? In table 1 below, describe the characteristics of particles for each phase of matter based on Model 1 above. Be specific.

| Table 1. | Solid | Liquid | Gas |
|-------------------------|-------|--------|-----|
| Depict the Particles | | | |
| Describe the Particles | | | |
| Motion of the Particles | | | |

States Of Matter Pogil Key

AN Whitehead



States Of Matter Pogil Key:

Mandy Mixes It Up With States of Matter Carole Marsh, 2008 Activities for students help them explore states of matter
CBSE Science Chapterwise Case Study Class 9 Priti Singhal, 2024-11-17 This book is structured to align with the latest syllabus and curriculum guidelines ensuring that the content is both relevant and rigorous Each chapter begins with a clear set of learning objectives providing a roadmap for students to understand what they will achieve by the end of the chapter We have included numerous diagrams illustrations and real life examples to make complex concepts more accessible and engaging *Growing with Science and Health 2 Teacher's Manual* 1st Ed. 1997 , Cyber Science 6 Tm' 2007 Ed. ,

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