Moles and calculations

Define moles: It is the amount of substance which is equal to the Avogadro's number of particles.

Avogadro's number or Avogadro's Constant = 6.02 x 10²⁰ particles Formulas for calculation of moles:

1. mole = mass

Mass of one mole

Mass of one mole = molar mass = Mr or Ar

2. mole of any gas = Volume of gas in cm³ Volume of gas in dm³

Volume of one mole of gas Volume of one mole of gas

Volume of one mole of any gas = molar gas volume = 24000 cm3 or 24 dm3

mole of any substance = volume in dm³ x concentration in mol dm³ (In the form of solution)

Volume in dm³ = Volume of solution in cm³

4. percentage yield = Experimental yield × 100

5. percentage purity = Mass of pure substance × 100

7. molecular formula = n x empirical formula

Where n = Empirical formula mass

Skill Practice Moles And Formula Practice

Jay Wackerly, Sarah Zingales, Michael Wentzel, Gautam Bhattacharyya, Brett McCollum

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