Name ______ Period _____ Date _____

Chapter 5: Thermal Energy

Lesson 1: Thermal Energy, Temperature, and Heat

Thermal energy - The sum of kinetic and potential energy of the particles that make up a material

- Thermal energy causes a phase change.
- Thermal energy describes the energy of the particles that make up a solid.
 liquid, or gas.

Thermal Energy - Kinetic

- Every solid, liquid, gas contains trillions of tiny particles that are in motion.
- Because these particles are in motion they all have kinetic energy.
- The <u>faster</u> the particles move, the more <u>kinetic energy</u> they have.
 - o Solids particles vibrate in place
 - Liquids particles move a little more freely
 - Gases particles move most freely and quickly

Thermal Energy - Potential

- Particles that make up <u>solids</u>, <u>liquids</u>, <u>and gases</u> also have <u>potential energy</u> which is the <u>distance between particles</u>.
- The greater the <u>distance</u> between <u>particles</u> the greater the <u>potential energy</u> of the <u>particles</u>.
 - o Solids Particles are held close together by attractive forces
 - Liquids <u>Particles are slightly farther apart</u>
 - o Gases Particles are much more spread out than in those in a solid

Heat - The movement of thermal energy from a warmer object to a cooler object
Why do your hands get warm when you hold a cup of coffee? The thermal energy
from the warm cup is moving to and heating up your cool hands.

What Are The Answers To Chapter Thermal Energy

Yijin Wang

What Are The Answers To Chapter Thermal Energy:

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