

Mark Scheme

Q1.

Q	Answer	Mark
	The only correct answer is B (CH_3CO^+) <i>A is incorrect because the fragment is not present in propanone</i> <i>C is incorrect because the fragment is not present in propanone</i> <i>D is incorrect because the fragment is not present in propanone</i>	1

Q2.

Question Number	Answer	Mark
(4)	The only correct answer is B (oxidising propan-1-ol to propanal) <i>A is incorrect because reducing an alcohol would produce an alkane</i> <i>C is incorrect because reducing propanal would produce propan-1-ol</i> <i>D is incorrect because oxidising propan-1-ol would produce propanal or propanoic acid</i>	(1)

Question Number	Answer	Mark
(5)	The only correct answer is A (propan-1-ol) <i>B is incorrect because propan-2-ol would not be expected to form a $^{\circ}\text{CH}_2\text{OH}$ fragment</i> <i>C is incorrect because propanal would not be expected to form a $^{\circ}\text{CH}_2\text{OH}$ fragment</i> <i>D is incorrect because propanone would not be expected to form a $^{\circ}\text{CH}_2\text{OH}$ fragment</i>	(1)

Question Number	Answer	Mark
(6)	The only correct answer is C (propanal) <i>A is incorrect because propan-1-ol would have a broad absorption at $3750\text{--}3200\text{ cm}^{-1}$ due to $-\text{OH}$</i> <i>B is incorrect because propan-2-ol would have a broad absorption at $3750\text{--}3200\text{ cm}^{-1}$ due to $-\text{OH}$</i> <i>D is incorrect because the absorption due to $\text{C}=\text{O}$ in propanone would be at $1720\text{--}1700\text{ cm}^{-1}$ and C-H stretching vibrations at $2775\text{--}2700\text{ cm}^{-1}$ would be absent</i>	(1)

Q3.

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Cameron Dunn



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