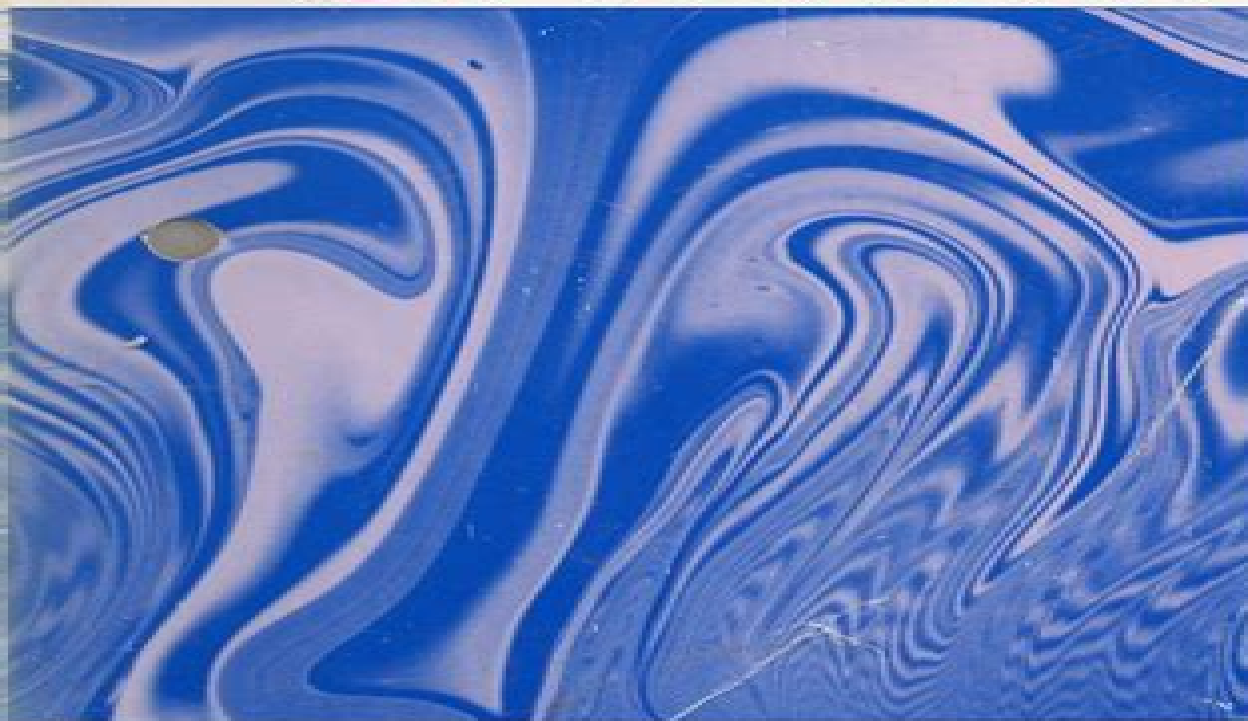


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often used as a synonym for scientific. Recent scholarship in history and philosophy of science has shown that it is not the experimental data Baconian orgy of quantification but rather the diversity plurality in a scientific discipline that contributes toward understanding objectivity. History of science shows that objectivity and subjectivity can be considered as the two poles of a continuum and this dualism leads to a conflict in understanding the evolving nature of objectivity. The history of objectivity is nothing less than the history of science itself and the evolving and varying forms of objectivity does not mean that one replaced the other in a sequence but rather each form supplements the others. This book is remarkable for its insistence that the philosophy of science and in particular that discipline's analysis of objectivity as the supposed hallmark of the scientific method is of direct value to teachers of science. Meticulously yet in a most readable way Mansoor Niaz looks at the way objectivity has been dealt with over the years in influential educational journals and in textbooks it's fascinating how certain perspectives fade while basic questions show no sign of going away. There are few books that take both philosophy and education seriously this one does. Roald Hoffmann, Cornell University chemist, writer and Nobel Laureate in Chemistry.

Fundamentals of Environmental and Toxicological Chemistry Stanley E. Manahan, 2013-02-25. Fundamentals of Environmental and Toxicological Chemistry Sustainable Science Fourth Edition covers university level environmental chemistry with toxicological chemistry integrated throughout the book. This new edition of a bestseller provides an updated text with an increased emphasis on sustainability and green chemistry. It is organized based on the five spheres of Earth's environment: 1 the hydrosphere (water), 2 the atmosphere (air), 3 the geosphere (solid Earth), 4 the biosphere (life) and 5 the anthrosphere (the part of the environment made and used by humans). The first chapter defines environmental chemistry and each of the five environmental spheres. The second chapter presents the basics of toxicological chemistry and its relationship to environmental chemistry. Subsequent chapters are grouped by sphere, beginning with the hydrosphere and its environmental chemistry (water pollution, sustainability and water as nature's most renewable resource). Chapters then describe the atmosphere (its structure and importance for protecting life on Earth, air pollutants and the sustainability of atmospheric quality). The author explains the nature of the geosphere and discusses soil for growing food as well as geosphere sustainability. He also describes the biosphere and its sustainability. The final sphere described is the anthrosphere. The text explains human influence on the environment including climate pollution in and by the anthrosphere and means of sustaining this sphere. It also discusses renewable nonpolluting energy and introduces workplace monitoring. For readers needing additional basic chemistry background, the book includes two chapters on general chemistry and organic chemistry. This updated edition includes three new chapters, new examples and figures and many new homework problems.

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