

Activity B:**Determining concentration**Get the Gizmo ready:

- Click **Reset**. Select **1.00 M NaOH** for the Burette and **Mystery H_2SO_4** for the Flask.
- Select **Bromthymol blue** for the Indicator.



Introduction: Adding a drop of strong acid or base into a neutralized solution is similar to adding a drop of strong acid or base to water—it causes an abrupt change in pH. By using an appropriate indicator, a chemist can tell when a solution is neutralized by monitoring its color.

Question: How is titration used to determine an unknown concentration?

1. **Measure:** Titrate the sulfuric acid analyte (H_2SO_4) with the sodium hydroxide titrant (NaOH).

How much 1.00 M NaOH is needed to neutralize the H_2SO_4 solution?

21.6 mL

2. **Interpret:** The balanced equation for the reaction of H_2SO_4 and NaOH is given at bottom right.

Based on this equation, how many moles of NaOH react with 1 mole of H_2SO_4 ?

2

3. **Manipulate:** Recall that molarity is equal to the number of moles of a substance dissolved in one liter of solution; molarity = moles ÷ volume.

- A. Write an equation for determining the number of moles of NaOH that are added to the flask based on [NaOH] and volume of NaOH titrant (mL NaOH):

Moles NaOH =

$$[NaOH] \times mL\ NaOH / 1,000\ mL$$

- B. Write a similar expression for the number of moles of H_2SO_4 in the flask based on [H_2SO_4] and the volume of H_2SO_4 (mL).

Moles H_2SO_4 =

$$[H_2SO_4] \times mL\ H_2SO_4 / 1,000\ mL$$

- C. Because there are twice as many moles of NaOH as moles of H_2SO_4 in this reaction, you can say:

$$\text{Moles NaOH} = 2 \cdot \text{Moles } H_2SO_4$$

Substitute your expressions from 3A and 3B into this equation and solve for [H_2SO_4]:

Titration Virtual Activity Answer Key

Izet Kapetanović

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Titration Virtual Activity Answer Key:

An Introduction to Medicinal Chemistry Graham L. Patrick, 2023 The market leader in medicinal chemistry clear supportive and practical It helps students to effortlessly make the link from theory to real life applications using practical and focused coverage alongside a package of supportive online resources *Teaching Chemistry - A Studybook* Ingo Eilks, Avi Hofstein, 2013-04-20 This book focuses on developing and updating prospective and practicing chemistry teachers pedagogical content knowledge The 11 chapters of the book discuss the most essential theories from general and science education and in the second part of each of the chapters apply the theory to examples from the chemistry classroom Key sentences tasks for self assessment and suggestions for further reading are also included The book is focused on many different issues a teacher of chemistry is concerned with The chapters provide contemporary discussions of the chemistry curriculum objectives and assessment motivation learning difficulties linguistic issues practical work student active pedagogies ICT informal learning continuous professional development and teaching chemistry in developing environments This book with contributions from many of the world's top experts in chemistry education is a major publication offering something that has not previously been available Within this single volume chemistry teachers teacher educators and prospective teachers will find information and advice relating to key issues in teaching such as the curriculum assessment and so forth but contextualised in terms of the specifics of teaching and learning of chemistry and drawing upon the extensive research in the field Moreover the book is written in a scholarly style with extensive citations to the literature thus providing an excellent starting point for teachers and research students undertaking scholarly studies in chemistry education whilst at the same time offering insight and practical advice to support the planning of effective chemistry teaching This book should be considered essential reading for those preparing for chemistry teaching and will be an important addition to the libraries of all concerned with chemical education Dr Keith S Taber University of Cambridge Editor Chemistry Education Research and Practice The highly regarded collection of authors in this book fills a critical void by providing an essential resource for teachers of chemistry to enhance pedagogical content knowledge for teaching modern chemistry Through clever orchestration of examples and theory and with carefully framed guiding questions the book equips teachers to act on the relevance of essential chemistry knowledge to navigate such challenges as context motivation to learn thinking activity language assessment and maintaining professional expertise If you are a secondary or post secondary teacher of chemistry this book will quickly become a favorite well thumbed resource Professor Hannah Sevan University of Massachusetts Boston

Plant Products for Antiviral Therapeutics Banasri Hazra, Anirban Mandal, Vijay Kumar Prajapati, Paul F. Moundipa, 2022-11-18 *Drug Discovery and Development* Izet Kapetanović, 2011-12-16 Drug discovery and development process aims to make available medications that are safe and effective in improving the length and quality of life and relieving pain and suffering However the process is very complex time consuming resource intensive requiring multi

disciplinary expertise and innovative approaches There is a growing urgency to identify and develop more effective efficient and expedient ways to bring safe and effective products to the market The drug discovery and development process relies on the utilization of relevant and robust tools methods models and validated biomarkers that are predictive of clinical effects in terms of diagnosis prevention therapy and prognosis There is a growing emphasis on translational research a bidirectional bench to the bedside approach in an effort to improve the process efficiency and the need for further innovations The authors in the book discuss the current and evolving state of drug discovery and development

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