

Experiment 2 - Gravimetric Analysis of a Chloride Salt (G)

Purpose:

Determining quantitative value of chloride in an unknown soluble salt by showing and explaining common methods in gravimetric analysis.

Theory:

- Reaction between silver ion and chloride ion to form silver chloride as an equation:



- AgCl solubility in water and the K_{sp} :

Silver chloride is very insoluble, so we can assume that the above reaction goes to completion.

$$K_{sp} = 1.6 \times 10^{-10}$$

- AgCl is precipitated with an acidic solution to reduce likelihood of acid ions interfering with results. These ions will form co-precipitates with silver in the acid, due to these co-precipitates, the results would be higher.
- To prevent the precipitate from going through the filtered paper, it is heated to coagulate it into a different texture that's easier to work with. Heating the precipitate makes it coagulate, therefore, it is harder for the precipitate to go through the filtered paper.
- Dry silver chloride decomposes into silver and chlorine in the presence of light:



- If photodecomposition takes place in air, results will be lower, since chlorine will be released in the form of gas. If photochemical decomposition occurs in the presence of excess silver ions in aqueous solution, it will result in higher results.

Unknown Salt Lab Report

SB Merriam

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'SB Merriam' text.

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English in Analytical Chemistry. Communicating about Methods & Techniques. Книга для студента Надежда Зорина, Александра Соболева, 2022-10-19 **Lab Manual for Investigating Chemistry** David Collins (Ph. D.), Matthew Johll, 2008-12-02 While many of the core labs from the first edition have been retained a renewed focus on the basics of chemistry and the scientific process create an even more detailed supplemental offering **Modern Experimental Chemistry** George W. Jr. Latimer, 2012-12-02 Modern Experimental Chemistry provides techniques of qualitative analysis that reinforce experiments on ionic equilibria This book includes the determination of water in hydrated salts identification of an organic compound after determining its molecular weight and nonaqueous titration of a salt of a weak acid The calculation of chemical stoichiometry calculation of thermodynamic properties by determining the change in equilibrium with temperature and chromium chemistry are also covered This compilation contains enough experiments for classes which have six hours of laboratory two 3 hour meetings per week to last two semesters This publication is intended for chemistry students as an introductory manual to chemistry laboratory **The Essential Lab Manual** Karen Timberlake, 2002-06-24 Drawing from the successful main Laboratory Manual the Essential Laboratory Manual includes twenty one experiments which have been revised and updated Suitable for a one or two term lab course **Lab Manual eBook for Criminalistics: Forensic Science, Crime, and Terrorism - 365-Day Access** James E. Girard, 2021-10-12 Lab Manual eBook for Criminalistics Forensic Science Crime and Terrorism is a digital only eBook lab manual with 365 day access This Lab Manual eBook consists of 12 related experiments created by James Girard and arranged by chapter It provides hands on practice to students allowing them to apply key concepts presented in the text or eBook **Laboratory Experiments for General Chemistry** Harold R. Hunt, Toby F. Block, 1994 *Lab Experiments in Introductory Chemistry* Phil Reedy, Donald J. Wink, Sharon Fetzer-Gislason, 2003-03-21 The manual contains laboratory experiments written specifically for the prep chem lab as well as for the general chemistry course Available as a complete manual or custom published at <http://custompub.whfreeman.com> **General, Organic, and Biological Chemistry Study Guide and Selected Solutions** Karen C. Timberlake, 2001-11 Keyed to the learning goals in the text this guide is designed to promote active learning through a variety of exercises with answers and mastery exams The guide also contains complete solutions to odd numbered problems **General Chemistry** Hunt, 1976-06-07 *Experiments in General Chemistry* Kenneth W. Whitten, Kenneth D. Gailey, 1981 *Chemical Investigations for Changing Times* C. Alton Hassell, John Hill, 2001 *U.S. Government Research Reports*, 1964 *Scientific and Technical Aerospace Reports*, 1995 Lists citations with abstracts for aerospace related reports obtained from world wide sources and announces documents that have recently been entered into the NASA Scientific and Technical Information Database *How to Write to Learn Science* Bob Tierney, John Dorroh, 2004 Make science an exhilarating process of discovery Through a wealth of creative write to learn strategies this book offers

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