

CHEMISTRY

Paper 5070/42
Alternative to Practical

Key messages

In calculations, candidates are advised to give their answers to more, rather than fewer, significant figures. Rounding down or rounding up to too few significant figures often leads to answers being given to an insufficient degree of accuracy. If in doubt, candidates should give their answers to a minimum of three significant figures.

Comments on specific questions

Question 1

- (a) Many candidates obviously recognised what the errors were, but found it difficult to express their answers in a clear manner. Common incorrect or insufficient answers included:-
- water in and water out were the wrong way round on the Liebig condenser
 - the thermometer was in the wrong position (without giving a positive statement about what the correct position should be)
 - the flask had a bung in it without reference to which flask the candidate was referring to. (A bung is essential on the distillation flask)
- (b) (i) This was answered well, although "fractionation column", "fractionalating column" or "fractional column" were often seen.
- (ii) The purpose of a fractionating column is separation. Answers occasionally focused on the high surface area of the glass beads rather than concentrating on the purpose of the apparatus.
- (iii) This was usually correct, although condenser was often incorrectly spelt.
- (iv) Some candidates indicated that water is the liquid which collects as the distillate as opposed to being used as a coolant. Others realised that the vapour was being cooled, although there was no reference to the effect of the cooling which was to turn the vapour into a liquid.
- (c) (i) (ii) A large majority of candidates answered both of these parts correctly.
- (d) Candidates should be aware that the majority of organic liquids are flammable and that direct heat from a flame should not be used. Alcohols were occasionally mentioned instead of alkanes. Several suitable alternative methods of heating were suggested by candidates, although uniform heating was often given as the reason instead of flammability.

Question 2

- (a) Many candidates were fully aware of how to carry out a test for nitrates and how to recognise a positive result. Heat was occasionally absent.
- (b) Most candidates gained partial credit for this question. In some cases evaporation to dryness was chosen. Washing the crystals was usually absent. There was no requirement to make the salt or to make the aqueous solution.
- (c) (i) This was answered very well, although a small number of candidates entered the readings the wrong way round.

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Peter Cann



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Table of Contents University Of Cambridge International Examinations Chemistry Past Papers

1. Understanding the eBook University Of Cambridge International Examinations Chemistry Past Papers
 - The Rise of Digital Reading University Of Cambridge International Examinations Chemistry Past Papers
 - Advantages of eBooks Over Traditional Books
2. Identifying University Of Cambridge International Examinations Chemistry Past Papers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an University Of Cambridge International Examinations Chemistry Past Papers
 - User-Friendly Interface
4. Exploring eBook Recommendations from University Of Cambridge International Examinations Chemistry Past Papers
 - Personalized Recommendations
 - University Of Cambridge International Examinations Chemistry Past Papers User Reviews and Ratings
 - University Of Cambridge International Examinations Chemistry Past Papers and Bestseller Lists

5. Accessing University Of Cambridge International Examinations Chemistry Past Papers Free and Paid eBooks
 - University Of Cambridge International Examinations Chemistry Past Papers Public Domain eBooks
 - University Of Cambridge International Examinations Chemistry Past Papers eBook Subscription Services
 - University Of Cambridge International Examinations Chemistry Past Papers Budget-Friendly Options
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 - ePub, PDF, MOBI, and More
 - University Of Cambridge International Examinations Chemistry Past Papers Compatibility with Devices
 - University Of Cambridge International Examinations Chemistry Past Papers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of University Of Cambridge International Examinations Chemistry Past Papers
 - Highlighting and Note-Taking University Of Cambridge International Examinations Chemistry Past Papers
 - Interactive Elements University Of Cambridge International Examinations Chemistry Past Papers
8. Staying Engaged with University Of Cambridge International Examinations Chemistry Past Papers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers University Of Cambridge International Examinations Chemistry Past Papers
9. Balancing eBooks and Physical Books University Of Cambridge International Examinations Chemistry Past Papers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection University Of Cambridge International Examinations Chemistry Past Papers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine University Of Cambridge International Examinations Chemistry Past Papers
 - Setting Reading Goals University Of Cambridge International Examinations Chemistry Past Papers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of University Of Cambridge International Examinations Chemistry Past Papers
 - Fact-Checking eBook Content of University Of Cambridge International Examinations Chemistry Past Papers
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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