

Titration Practice Worksheet

Find the requested quantities in the following problems:

- 1) If it takes 54 mL of 0.1 M NaOH to neutralize 125 mL of an HCl solution, what is the concentration of the HCl?

$$\begin{aligned} M_1 &= 0.1 \text{ M} \\ V_1 &= 54 \text{ mL} \\ M_2 &= ? \\ V_2 &= 125 \text{ mL} \end{aligned}$$

$$\begin{aligned} M_1 V_1 &= M_2 V_2 \\ (0.1 \text{ M})(54 \text{ mL}) &= (125 \text{ mL}) M_2 \\ M_2 &= \frac{(0.1 \text{ M})(54 \text{ mL})}{125 \text{ mL}} = 0.0432 \text{ M} = 4.32 \times 10^{-2} \text{ M} \end{aligned}$$

The concentration of HCl is $4.32 \times 10^{-2} \text{ M}$.

- 2) If it takes 25 mL of 0.05 M HCl to neutralize 345 mL of NaOH solution, what is the concentration of the NaOH solution?

$$\begin{aligned} M_1 &= 0.05 \text{ M} \\ V_1 &= 25 \text{ mL} \\ M_2 &= ? \\ V_2 &= 345 \text{ mL} \end{aligned}$$

$$\begin{aligned} M_1 V_1 &= M_2 V_2 \\ (0.05 \text{ M})(25 \text{ mL}) &= (345 \text{ mL}) M_2 \\ M_2 &= \frac{(0.05 \text{ M})(25 \text{ mL})}{345 \text{ mL}} = 0.00362 \text{ M} = 3.62 \times 10^{-3} \text{ M} \end{aligned}$$

The concentration of NaOH is $3.62 \times 10^{-3} \text{ M}$.

- 3) If it takes 50 mL of 0.5 M KOH solution to completely neutralize 125 mL of sulfuric acid solution (H_2SO_4), what is the concentration of the H_2SO_4 solution?



$$n = MV$$

$$n = (0.5 \text{ M})(50 \text{ mL}) = 25 \text{ mol OH}^-$$

2 times $[\text{OH}^-]$ is needed

to neutralize $[\text{H}^+]$, $\rightarrow [\text{H}^+] = \frac{1}{2} [\text{OH}^-]$

$$M = \frac{12.5 \text{ mol}}{125 \text{ mL}} = \boxed{0.10 \text{ M}}$$

- 4) Can I titrate a solution of unknown concentration and still get a meaningful answer? Explain your answer in a few sentences.

Titration cannot be done without molarity of at least one of the substances. In order to solve $M_1 V_1 = M_2 V_2$, both molarities cannot be unknown.

- 5) Explain the difference between an endpoint and equivalence point in a titration.

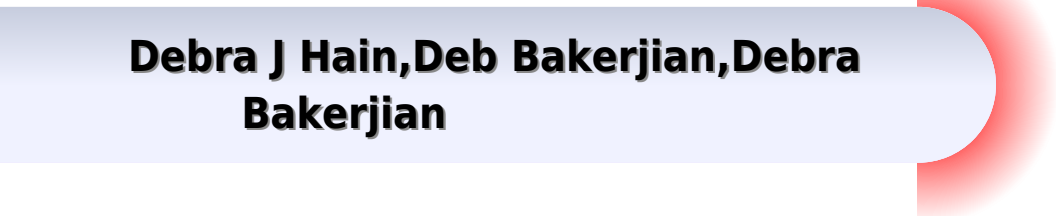
Endpoint: titration can be stopped.

Equivalence point: when $[\text{H}^+]$ and $[\text{OH}^-]$ are neutralized.

The effectiveness of a titration is measured by the closeness of the endpoint to the equivalence point.

Titration Nursing Math Problems With Key

**Debra J Hain, Deb Bakerjian, Debra
Bakerjian**



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Textbook of Adult-Gerontology Primary Care Nursing Debra J Hain, Deb Bakerjian, Debra Bakerjian, 2022-02-21 I was thrilled to see content that focuses on quality improvement patient safety interprofessional collaboration care coordination and other content that supports the role of the AGNP as a clinical leader and change agent The authors give these topics the attention that they deserve with clear insightful guidance and importantly the evidence base The chapters that address roles including during disasters settings of care billing and medication use address salient issues that will help the fledgling AGNP to hit the ground running and the seasoned AGNP to keep current Marie Boltz PhD GNP BC FGSA FAAN Elouise Ross Eberly and Robert Eberly Endowed Professor Toss and Carol Nese College of Nursing Penn State University From the Foreword Written for Adult Gerontology Primary Care Nurse Practitioners faculty and students this primary text encompasses the full scope of AGNP primary care practice across multiple healthcare settings including telehealth The text emphasizes the best available evidence to promote person centered care quality improvement of care interprofessional collaboration and reducing healthcare costs The text delivers timely information about current healthcare initiatives in the U S including care coordination across the healthcare continuum interprofessional collaboration and accountable care organizations Disease focused chapters contain general and specific population based assessment and interprofessional care strategies to both common and complex health issues They offer consistent content on emergencies relevant social determinants of health and

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O escolar Débora Marques de Miranda,Leandro Fernandes Malloy-Diniz,2020-07-20 O escolar oferece um panorama atual sobre os anos de vida escolar com os impactos que a escola pode propiciar para o bem e para o mal das crianças dessa idade O escolar apresenta o desenvolvimento típico da criança em idade escolar os mecanismos para tornar esse processo saudável os sintomas de situações de alerta e os principais fatores de risco observados ao desenvolvimento saudável

Reafirmamos que o conteúdo desta obra também é resultante da integração dos efeitos promovidos pelos ambientes social, biológico e psicológico demonstrando a importância de abordagem integral. Exemplificamos com o clássico da literatura e abordamos novos problemas da modernidade cuja literatura esteja crescente como os tempos de exposição a telas e o uso de jogos nessa faixa etária. Este livro condensa evidências e experiências de autores engajados no desenvolvimento humano e na otimização de potenciais de capital mental. Parte da coleção Desenvolvimento Humano no Ciclo da Vida, esta obra busca apresentar as etapas da vida com suas particularidades, mas com a perspectiva contínua de que todas as fases devem ser compreendidas e trabalhadas de forma a otimizar os potenciais para o indivíduo e para a sociedade. Profissionais experientes tecem comentários sobre problemas atuais e informam dentro do melhor que a literatura científica sugere como abordar a criança dessa faixa etária de forma a evitar déficits e aprimorar a trajetória de seu desenvolvimento, otimizando o capital mental do indivíduo e da sociedade. *World Who's who in Science* Allen G. Debus, 1968

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