

Activity B:**Determining concentration**Get the Gizmo ready:

- Click **Reset**. Select **1.00 M NaOH** for the Burette and **Mystery H_2SO_4** for the Flask.
- Select **Bromthymol blue** for the Indicator.



Introduction: Adding a drop of strong acid or base into a neutralized solution is similar to adding a drop of strong acid or base to water—it causes an abrupt change in pH. By using an appropriate indicator, a chemist can tell when a solution is neutralized by monitoring its color.

Question: How is titration used to determine an unknown concentration?

1. **Measure:** Titrate the sulfuric acid analyte (H_2SO_4) with the sodium hydroxide titrant (NaOH).

How much 1.00 M NaOH is needed to neutralize the H_2SO_4 solution?

21.6 mL

2. **Interpret:** The balanced equation for the reaction of H_2SO_4 and NaOH is given at bottom right.

Based on this equation, how many moles of NaOH react with 1 mole of H_2SO_4 ?

2

3. **Manipulate:** Recall that molarity is equal to the number of moles of a substance dissolved in one liter of solution; molarity = moles ÷ volume.

- A. Write an equation for determining the number of moles of NaOH that are added to the flask based on [NaOH] and volume of NaOH titrant (mL NaOH):

Moles NaOH =

$$[NaOH] \times mL\ NaOH / 1,000\ mL$$

- B. Write a similar expression for the number of moles of H_2SO_4 in the flask based on [H_2SO_4] and the volume of H_2SO_4 (mL).

Moles H_2SO_4 =

$$[H_2SO_4] \times mL\ H_2SO_4 / 1,000\ mL$$

- C. Because there are twice as many moles of NaOH as moles of H_2SO_4 in this reaction, you can say:

$$\text{Moles NaOH} = 2 \cdot \text{Moles } H_2SO_4$$

Substitute your expressions from 3A and 3B into this equation and solve for [H_2SO_4]:

Titration Virtual Activity 1 Answer Key

**Andrew Burrows, Andrew
Parsons, Gwen Pilling, Gareth Price**

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Yue, 2021-09-19 This book presents the general objective of the REV2021 conference which is to contribute and discuss fundamentals applications and experiences in the field of Online and Remote Engineering Virtual Instrumentation and other related new technologies like Cross Reality Data Science Big Data Internet of Things Industrial Internet of Things Industry 4 0 Cyber Security and M2M Smart Objects Nowadays online technologies are the core of most fields of engineering and the whole society and are inseparably connected for example with Internet of Things Industry 4 0 Industrial Internet of Things Cloud Technologies Data Science Cross Mixed Reality Remote Working Environments Online Biomedical Engineering to name only a few Since the first REV conference in 2004 we tried to focus on the upcoming use of the Internet for engineering tasks and the opportunities as well as challenges around it In a globally connected world the interest in online collaboration teleworking remote services and other digital working environments is rapidly increasing Another objective of the conference is to discuss guidelines and new concepts for engineering education in higher and vocational education institutions including emerging technologies in learning MOOCs MOOLs and Open Resources REV2021 on Online Engineering and Society 4 0 was the 17th in a series of annual events concerning the area of Remote Engineering and Virtual Instrumentation It has been organized in cooperation with the International Engineering and Technology Institute IETI as an online event from February 24 to 26 2021 *AP Chemistry Premium, 2024: 6 Practice Tests + Comprehensive Review + Online Practice* Neil D.

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discussion of these electrodes is not arranged according to the species detected but rather according to the kind of construction since from this view point characteristic properties are much the same and handling procedures need only be described once for an entire series of similar electrodes The fourth section discusses amplifiers Here the problems of high ohmic EMF measurements such as noise level insulation static charging and ground loops are discussed The fifth section is devoted to the various evaluation methods Here a few schemes and examples are provided to indicate optimum practical procedures and the accuracies attainable with the various methods are discussed The last section describes special set ups such as clinical flow thru cells microelectrodes for measuring intracellular ionic activities industrial on line techniques and continuous environmental protection monitors

*Chemistry*³ Andrew Burrows, Andrew Parsons, Gwen Pilling, Gareth Price, 2013-03-21 New to this Edition

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