

FORMATIVE ASSESSMENT: PROPERTIES OF WATER/pH scale

Name: _____ Teacher: _____ Date: _____

Learning Goals:

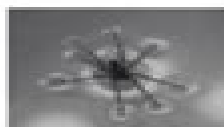
Students will be able to:

Understand how water supports life.

Understand the pH scale and how relates to acids and bases.

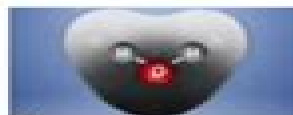
Score 2:

- Water droplets form by water molecules being attracted to each other. What term best explains this behavior?
 - adhesion
 - cohesion
 - solubility
 - magnetism
- Which property of water allows water to stick to the sides of the straw?
 - adhesion
 - cohesion
 - heat capacity
 - solubility
- Which property of water allows bugs to walk on water?
 - adhesion
 - heat capacity
 - surface tension
 - solubility
- Water takes _____ heat to change to its gaseous state.
 - large amounts of
 - small amounts of
 - no
- What property of water allows sweat to cool the body & coastal cities to have more moderate temperatures than inland cities?
 - High heat capacity
 - Neutral pH of 7
 - Surface tension and capillary action
 - Universal solvent
- A solution with a pH of 3 is:
 - acidic
 - basic
 - neutral
- A solution with a pH of 8 is:
 - acidic
 - basic
 - neutral



Score 3:

- Water is a polar molecule because
 - It shares electrons equally between oxygen and hydrogen
 - One side of the molecule has a slight positive charge and the other side has a slight negative charge
 - It freezes at zero-degrees Celsius
 - It forms hydrogen bonds with other water molecules
- Which side of the water molecule has a slight negative charge?
 - oxygen side
 - hydrogen side



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