

# Graphing Periodic Trends Activity

Name: \_\_\_\_\_ Period: \_\_\_\_\_

Use the information in these tables to complete the graph as described below

| Symbol | Atomic Radius (Picometers) | First Ionization energy (kJoules/mole) | Electronegativity (4-point scale) |
|--------|----------------------------|----------------------------------------|-----------------------------------|
| H      | 31                         | 1312                                   | 2.1                               |
| Li     | 128                        | 520                                    | 1.0                               |
| Na     | 166                        | 496                                    | 0.9                               |
| K      | 203                        | 410                                    | 0.8                               |
| Rb     | 220                        | 403                                    | 0.8                               |
| Cs     | 244                        | 376                                    | 0.7                               |

| Symbol | Atomic Radius (Picometers) | First Ionization energy (kJoules/mole) | Electronegativity (4-point scale) |
|--------|----------------------------|----------------------------------------|-----------------------------------|
| Na     | 166                        | 496                                    | 0.9                               |
| Mg     | 141                        | 738                                    | 1.2                               |
| Al     | 121                        | 578                                    | 1.5                               |
| Si     | 111                        | 787                                    | 1.8                               |
| P      | 107                        | 1012                                   | 2.1                               |
| S      | 105                        | 1000                                   | 2.5                               |
| Cl     | 102                        | 1251                                   | 3.0                               |
| Ar     | 106                        | 1521                                   | ---                               |

1. Using colored pencils or pen, list under the symbol (in this order!) the Atomic Radius, First Ionization Energy, and Electronegativity.
2. Use a different color for each property. Example: Write all of the Atomic radius values in red, all of the First ionization energies in green, and all of the Electronegativities in blue. Of course, you can pick any colors that are available, as long as you are consistent.
3. Observe the trends in each property as you go down the Alkali metal group, and as you go across Period 3.
4. Write out each of the statements written on the opposite side, completing each statement with the observed trend (increase or decrease)

|                         |                                                                                                                                                                                          |    |    |   |   |    |    |
|-------------------------|------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|----|----|---|---|----|----|
| H                       | <p>Across a period (→) atomic radius tends to _____.</p> <p>Across a period (→) first ionization energy tends to _____.</p> <p>Across a period (→) electronegativity tends to _____.</p> |    |    |   |   |    |    |
| Li                      |                                                                                                                                                                                          |    |    |   |   |    |    |
| Na<br>166<br>496<br>0.9 | Mg                                                                                                                                                                                       | Al | Si | P | S | Cl | Ar |
| K                       | <p>Down a Group (↓) atomic radius tends to _____.</p> <p>Down a Group (↓) first ionization energy tends to _____.</p> <p>Down a Group (↓) electronegativity tends to _____.</p>          |    |    |   |   |    |    |
| Rb                      |                                                                                                                                                                                          |    |    |   |   |    |    |
| Cs                      |                                                                                                                                                                                          |    |    |   |   |    |    |

# Skill Sheet Periodic Trends

**Gray, Laura E., Dunn, Shernette D.**

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author names.

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The surge in enrollment for online courses is continually increasing However beneath the convenience lies a challenge that demands a resolution Educators administrators and instructional designers must ensure that the human element is not lost in the virtual corridors of learning Students with diverse backgrounds and learning needs require more than a simple virtual classroom Research reveals a concerning trend high attrition rates in online courses often attributed to a lack of engagement and insufficient human interaction To reverse this trend deliberate measures must be taken to humanize the online learning environment This book provides several solutions offering an array of tools and strategies to promote engagement and infuse the human touch into online spaces To confront this multi layered challenge it becomes paramount to undertake deliberate measures aimed at humanizing the online learning environment Humanizing Online Teaching and Learning in Higher Education steps forward as a guide offering an extensive array of tools and strategies meticulously crafted to foster student engagement and infuse the essential human touch into the digital educational landscape     Daily Graphic Ransford Tetteh, 2014-03-07     *Bulletin of the United States Bureau of Labor Statistics* , 1980     **Medical Infrastructure** United States. Congress. House. Committee on Armed Services. Subcommittee on Military Personnel, 2010

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