Math Skills

Specific Heat

How much energy must be transferred as heat to 200 kg of water in a bathtub to raise the water's temperature from 25 °C to 37 °C?

1. List the given and the unknown values.

Given:
$$\Delta T = 37 \,^{\circ}\text{C} - 25 \,^{\circ}\text{C} = 12 \,^{\circ}\text{C} = 12 \,^{\circ}\text{K}$$

 $\Delta T = 12 \,^{\circ}\text{K}$

$$m = 200 \text{ kg}$$

$$c = 4,186 \text{ J/kg} \cdot \text{K}$$

Unknown: energy = ? J









Skills Math Skills Specific Heat

AW Rasmussen

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