

States of Matter

Review and Reinforce

1. The bonds of attraction would be relaxed and so a liquid, and the vapors so a gas.
2. The solid would lose the water vapor because the particles in a solid are bound tightly in place. The vapor of the liquid will be able to spread out into a gas-like form; the particles in a liquid have spread out little. The vapor of the liquid would continue to change and spread outwards because the particles in a gas move apart in many different directions.
3. Since the bonds of water in solid ice volume would stay the same. The volume of the liquid will be less than the solid. The volume of the vapors would expand to fill the size of the larger container because the particles would spread apart.
4. Matter with a definite volume and a definite shape.
5. Matter with a definite volume but no shape will be a liquid.
6. Matter with a definite shape or volume.
7. The molecules of a liquid in motion.
8. It refers to where the particles are not arranged in a regular pattern.
9. It refers to where the particles are arranged in a regular repeating pattern.
10. A substance that flows.
11. The bonds of an ionic solid among the molecules of a liquid that bring surface molecules closer together.

Changes of State

Review and Reinforce

1. Gases, liquids
2. Gases, liquids
3. Gases, liquids
4. Gases, liquids
5. Gases, liquids
6. boiling point
7. condensation
8. evaporation
9. boiling
10. evaporation
11. melting
12. freezing
13. melting point
14. sublimation

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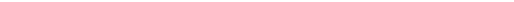
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