

FORMULAS FOR IONIC COMPOUNDS

1. Write the chemical formula for the ionic compounds made from the following elements.

Elements	metal ion	non-metal ion	Chemical Formula	Name
potassium and iodine	K^{+1}	I^{-1}	KI	Potassium Iodide
barium and chlorine	Ba^{2+}	Cl^{-1}	$BaCl_2$	Barium Chloride
sodium and oxygen	Na^{+1}	O^{2-}	Na_2O	sodium Oxide
aluminum and nitrogen	Al^{3+}	N^{3-}	AlN	Aluminum Nitride
lithium and sulfur	Li^{+1}	S^{2-}	Li_2S	Lithium Sulfide
magnesium and oxygen	Mg^{+2}	O^{2-}	MgO	Magnesium Oxide
calcium and chlorine	Ca^{2+}	Cl^{-1}	$CaCl_2$	calcium chloride
boron and chlorine	B^{3+}	Cl^{-1}	BCl_3	boron chloride
beryllium and sulfur	Be^{2+}	S^{2-}	BeS	beryllium sulfide

2. Complete the chart below and write the ionic formulas to complete it. A few have been done for you.

	F	Cl	O^{2-}	S^{2-}	N^{3-}	P^{3-}
Li^{+}	LiF	LiCl	Li_2O	Li_2S	Li_3N	Li_3P
Na^{+}	NaF	NaCl	Na ₂ O	Na_2S	Na_3N	Na_3P
K^{+}	KF	KCl	K_2O	K_2S	K_3N	K_3P
Be^{2+}	BeF_2	$BeCl_2$	BeO	BeS	Be_3S_2	Be_3P_2
Mg^{2+}	MgF_2	$MgCl_2$	MgO	MgS	Mg_3N_2	Mg_3P_2
Ca^{2+}	CaF_2	$CaCl_2$	CaO	CaS	Ca_3N_2	Ca_3P_2
Al^{3+}	AlF_3	$AlCl_3$	Al_2O_3	Al_2S_3	AlN	AlP
Ga^{3+}	GaF_3	$GaCl_3$	Ga_2O_3	Ga_2S_3	GaN	GaP
B^{3+}	BF_3	BCl_3	B_2O_3	B_2S_3	BN	BP

Skills Worksheet Math Skills Writing Ionic Formulas Answers

David C. McMillin



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Science Spectrum Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2003-03 *Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science*, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction Carolina Science and Math Carolina Biological Supply Company, 2003 **Chemistry** McGraw-Hill Staff, 2001-05-01 **Chemistry Math Concepts** David C. McMillin, 2001 **Essential Math for Chemistry Students** David W. Ball, 1996 Focuses on the skills needed to survive in general chemistry with worked examples showing how these skills translate into successful chemical problem solving An ideal tool for students lacking in confidence or competency in the essential math skills required for general chemistry

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