

STATES OF MATTER

- All substances are made of matter. Matter is all the stuff around you – everything you can touch, see or smell. Anything that has **mass** (weight) and **volume** (occupies space) is matter. **Atom**:- is the smallest particle of matter. Everything is made up of atoms.
- The three states of matter are solid, liquid and gas.
- Solids, liquids and gases are all made of tiny particles. This is called the **particle theory**.
- The particles are attracted to each other. Some particles are attracted strongly to each other and others weakly.
- The particles move around. They are described as having kinetic energy. The kinetic energy of the particles increases with temperature.

Solids

Particles of solids are tightly packed, are very close together in a fixed arrangement.

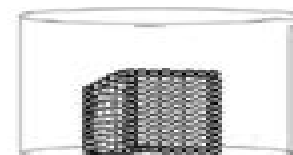
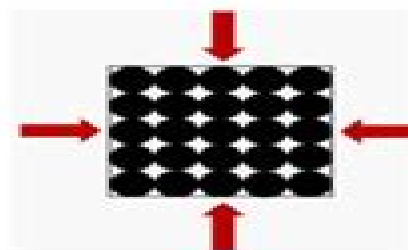
There are very strong forces of attraction between the particles holding them together.

There are very small spaces between the particles – high density.

This makes solids hard (usually) and they **cannot** be compressed.

Particles have a small amount of energy, particles vibrate but do not move around.

Solids have a definite shape and a definite volume.



Liquids

Particles of liquids are tightly packed, but are far enough apart to slide over one another.

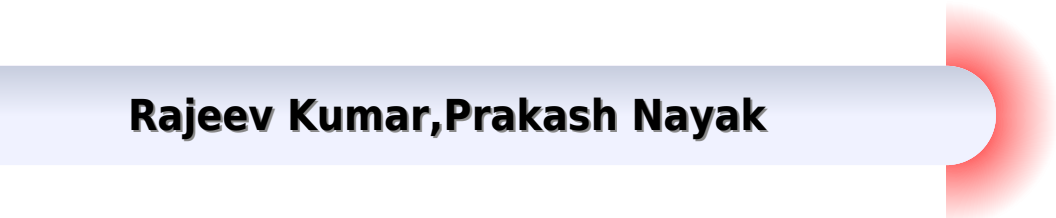
There are strong forces of attraction between the particles holding them together but not in an orderly arrangement.

There are small spaces between the particles – medium density.



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Rajeev Kumar,Prakash Nayak



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