

## Periodic Table Named-Groups Worksheet – 2

**DIRECTIONS:** Use the Periodic Table shown below to answer each question, using the letters in each "box" to identify the required element(s).

The diagram shows a crossword puzzle grid with the following letters placed in specific cells:

- Row 1:** A (1,1), B (15,1)
- Row 2:** C (1,2), D (10,2), E (12,2)
- Row 3:** F (2,3), G (6,3), H (9,3), I (11,3), K (14,3)
- Row 4:** J (1,4), L (2,4), M (5,4), N (8,4), O (12,4), P (14,4)
- Row 5:** Q (1,5), R (2,5), S (5,5), T (6,5), U (7,5)
- Row 6:** V (1,6), W (8,6), X (1,7), Y (8,7)

Arrows point from the letter 'R' in cell (2,5) to the letters 'V' in cell (1,6) and 'X' in cell (1,7).

List elements in *alphabetical order* when more than 1 element answers a question.

- \_\_\_\_\_ 1) Element **A** is in a box all by itself because: **A)** it is really a Noble Gas **B)** it is a gas **C)** it is NOT a gas **D)** it does NOT share the properties of the other elements in its group **E)** it is a liquid
- \_\_\_\_\_ 2) Which of these elements is/are highly reactive? **A)** "C" **B)** "E" **C)** "K" **D)** A, B, and C **E)** A and B only
- \_\_\_\_\_ 3) What is the letter of the element whose atomic number is closest to the element labeled "L"?
- \_\_\_\_\_ 4) Which letter labels the halogen element with the lowest atomic number?
- \_\_\_\_\_ 5) Which of these elements is the least reactive? **A)** "J" **B)** "Y" **C)** "A" **D)** "O" **E)** "B"
- \_\_\_\_\_ 6) Which of these elements is/are NOT transition elements/metals (list ALL that are not transition elements)? **A)** "C" **B)** "N" **C)** "K" **D)** "W" **E)** "Y"

continued on other side

# Skills Worksheet Reinforcement Periodic Table Answer Key

**Mike Jess**



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