EASEL PERIODIC TABLE QUIZ

Multiple Choice: Select the most correct answer. (15 K) at room temperature. I. Metals can be A) liquid only. B) solid only C) solid or liquid. D's solid, liquid, or gas E) liquid or gas. Most of the elements on the periodic table are All masses: B) pronumetals: Ch metalloids E) metals: D) liquids: 3. Which nonmetal exists as a diatomic solid? A) brooming B) antimony C) phosphorus D) inding E) boron-4. In which set of elements would all members be expected to have very similar chemical properties? C. Na. Mg. K. D) S. Se, St. E) Ne. Na. Mg A) O, S, Se B) N. O. F. 5. Which one of the following is not true about the alkali metals? A) They are low density solids at room temperature B) They all readily form ions with a +1 charge. C) They all have 2 electrons in their valence shells. D) They are very reactive elements. E) They have the lowest first ionization energies of the elements. is a unique element and does not truly belong to any family. A) Nitrogen B) Radium C) Hydrogen D) Uranium E) Helium 7. All of the halogens ______ A) exist under ambient conditions as diatomic gases B) tend to form positive ions of several different charges. C) tend to form negative ions of several different charges D) exhibit metallic character E) form sales with alkali metals with the formula MX. 8. In nature, the noble gases exist as: A) monatomic gaseous atoms B) the gaseous fluorides C) the sulfides D) alkali metal salts E) solids in rocks and in minerals. and the columns are called In the periodic table, the rows are called __ A) octaves, groups B) staffs, families C) periods, groups D) cogeners, families E) rows, groups 10. Which group in the periodic table contains only nonmetals? B) Group 6 D) Group 16 A) Group 1 Ch Caroun 10 E4 Geouge 18:

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