

# TEACHING TRANSPARENCY 22

Microscopic Chapter 2  
Section 1.1

## Formation of ions

- Q 1. What are cations in the following context?  
Electron deficient
- Q 2. How do cations appear in the last table of the last experiment (anion)? Chlorine (anion)  
positive electron deficient cation
- Q 3. What is the reason for the loss of electron in cations? Electron deficiency, positive (anion) is gained.  
Electron deficiency
- Q 4. What kind of change does the formation take place in cations? What is the reason for this?  
Electron deficient cation
- Q 5. What kind of change does the formation take place in anion? What is the reason for this?  
Electron deficient cation
- Q 6. In the above chemical combination of forming cations, why is the loss of electron? What is the reason for this?  
Electron deficiency, loss of electron
- Q 7. In the above chemical combination of forming cations, why is the loss of electron? What is the reason for this?  
Electron deficiency, loss of electron
- Q 8. In the above chemical combination of forming cations, why is the loss of electron? What is the reason for this?  
Electron deficiency, loss of electron
- Q 9. In the above chemical combination of forming cations, why is the loss of electron? What is the reason for this?  
Electron deficiency, loss of electron
- Q 10. What is the chemical combination of the loss of electron? What is the reason for this?  
Electron deficiency, loss of electron
- Q 11. What is the chemical combination of the loss of electron? What is the reason for this?  
Electron deficiency, loss of electron

# Teaching Transparency Formation Of Ions Answer Guide

**RJ Shavelson**



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