

The Periodic Table and Periodic Law

Section 6.1 Development of the Modern Periodic Table

In your textbook, read about the history of the periodic table's development.

Use each of the terms below just once to complete the passage.

octaves	atomic mass	atomic number	nine
elements	properties	Henry Moseley	eight
protons	periodic law	Dmitri Mendeleev	accepted

The table below was developed by John Newlands and is based on a relationship called the law of (1) _____. According to this law, the properties of the elements repeated every (2) _____ elements. Thus, for example, element two and element (3) _____ have similar properties. The law of octaves did not work for all the known elements and was not generally (4) _____.

1	2	3	4	5	6	7
H	Li	G	Bo	C	N	O
8	9	10	11	12	13	14
F	Na	Mg	Al	Si	P	S

The first periodic table is mostly credited to (5) _____. In his table, the elements were arranged according to increasing (6) _____. One important result of this table was that the existence and properties of undiscovered (7) _____ could be predicted.

The element in the modern periodic table are arranged according to increasing (8) _____, as a result of the work of (9) _____. This arrangement is based on number of (10) _____ in the nucleus of an atom of the element. The modern form of the periodic table results in the (11) _____, which states that when elements are arranged according to increasing atomic number, there is a periodic repetition of their chemical and physical (12) _____.

Teaching Transparency Worksheet The Periodic Table Answers

Linda Trombley



Teaching Transparency Worksheet The Periodic Table Answers:

Periodic Table Cindy Blobaum, 2005 What do chocolate chip cookies chemistry and logic have in common They are the basis for a unit that lets students become actively engaged in discovering the arrangement of the periodic table This learning activity takes the periodic table out of the static presentation usually associated with textbooks and chemistry courses and interjects an element of discovery The two activities in this unit provide students with information that they have to arrange in organized charts In the process of creating the arrangements students will be involved in problem solving and will gain an appreciation for the scientific process of exploration and verification This dynamic unit meets national science standards in seven teaching and content areas Bring the periodic table to life with this hands on minds on unit Book jacket **Chemistry in the Community** American Chemical Society, 2006-01-31 This laboratory based text centres itself around decision making activities where students apply their chemistry knowledge to realistic situations This fifth edition includes more photographs new drawings and new design *Science Spectrum* Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2003-03

Atoms, Molecules & Elements: The Periodic Table Gr. 5-8 George Graybill, 2015-10-01 This is the chapter slice The Periodic Table from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives **Elements and the Periodic Table, Grades 5 - 12** Theodore A. Abbgy, 2001-01-01 This informative classroom supplement is a great introduction to the periodic table explored in sequential form It includes activities transparency masters a teacher s guide an element game quizzes tests rubrics and answer keys Unit topics include discovering what elements are the uses of the elements element symbols periodic table organization and more Mark Twain Media Publishing Company specializes in providing captivating supplemental books and decorative resources to complement middle and upper grade classrooms Designed by leading educators the product line covers a range of subjects including mathematics sciences language arts social studies history government fine arts and character Mark Twain Media also provides innovative classroom solutions for bulletin boards and interactive whiteboards Since 1977 Mark Twain Media has remained a reliable source for a wide variety of engaging classroom resources **Atoms, Molecules & Elements: Patterns In the Periodic Table Gr. 5-8** George Graybill, 2015-10-01 This is the chapter slice Patterns In the Periodic Table from the full lesson plan

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of chemical behavior Students will quickly learn these three simple concepts and apply them to an astonishing number of chemical phenomena Within a few days they will accurately predict and explain the behavior of the elements covered in introductory chemistry courses These concepts will form a strong foundation for all the chemistry you ll teach in high school Your students will gain a level of fluency that is not possible using other approaches As teachers and parents know personal competency is the only reliable foundation for building self esteem and confidence Compatible with the emerging consensus on the nature of science Works with not against students natural cognitive growth Integrates language with learning

Facilitating Conceptual Change in Students' Understanding of the Periodic Table Mansoor Niaz,Marniev

Luiggi,2013-07-25 This book is about how students are taught the periodic table It reviews aspects of the periodic table s development using the history and philosophy of science The teaching method presented in this book is ideal for teaching the subject in high school and at introductory university level Chemistry students taught in this new experimental way are compared with those taught in the traditional way and the author describes how tests found more conceptual responses from the experimental group than the control group The historical aspects of importance to this teaching method are the role of the Karlsruhe Congress of 1860 the accommodation of the chemical elements in the periodic table prediction of elements that were discovered later corrections of atomic weights periodicity in the periodic table as a function of the atomic theory and the accommodation of argon The experimental group of students participated in various activities including discussion of various aspects related to the history and philosophy of science construction of concept maps and their evaluation by the students PowerPoint presentations and interviews with volunteer students

Techniques in Teaching the Periodic Table

Randal L. Musch,1994

Laminated Color Periodic Table and Formula Sheet for Chemistry, Biochemistry, and

Physics Laurence Lavelle,2012-07-17 Excellent presentation of the Periodic Table Visually appealing layout of Symbol Name Atomic Number Atomic Weight and Electron Configuration for each element

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