

Cracking the Periodic Table Code

Why aren't the elements listed in alphabetical order?

Why?

As charts go, the periodic table is a bit odd. It's not square. Large portions of the table appear to be missing at the top. It's not organized alphabetically so elements can be found easily. But to a chemist, the periodic table is a very powerful tool. The periodic table is organized by properties, both chemical and physical. Those properties relate to the electronic structure of the atoms of each element. In today's activity, you are going to study how the ground state electron configurations and structure of atoms are related to the shape and organization of the periodic table.

Blank Periodic Table

A blank periodic table grid with 7 rows and 18 columns. The grid is divided into s, p, d, and f blocks. The label $3p^3$ is located in the third row, eighth column, which corresponds to the element Nitrogen (N). The grid is intended for students to write the electron configuration for the element in that position.

1. Circle the number of the team your group has been assigned.

Team 1: H, Li, Na, K, Rb, Cs

Team 2: Fe, Co, Ni, Ru, Rh, Pd

Team 3: Ne, Ar, Kr, Xe, Rn

Team 4: F, Cl, Br, I, At

Team 5: Be, Mg, Ca, Sr, Ba

Team 6: N, P, As, Sb, Bi

Team 7: Ti, V, Cr, Zr, Nb, Mo

Team B: honors only Er, Tm, Yb, Fm, Md, No

Team 9: honors only Sm, Tb, Ho, Pu, Bk, Es

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ERDA Energy Research Abstracts United States. Energy Research and Development Administration, 1976-05

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