



## Student Exploration: Chemical Equations Activity A & B

**Directions:** Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

**Prior Knowledge Questions** (Do these BEFORE using the Gizmo.)

1. A candle is placed on one pan of a balance, and equal weight is placed on the other pan. What would happen if you lit up the candle and waited for a while?

The candle would eventually burn, thus losing mass and becoming lighter. The pan with the other object would lower as the candle burns.

2. Suppose the candle was placed in a large, sealed jar that allowed it to burn for several minutes before running out of oxygen. The candle and jar are balanced by equal weight. In this situation, what would happen if you lit up the candle and waited?

I assume the balance would stay level because the carbon dioxide is trapped within the jar, causing it to retain all of its mass despite the candle burning.

### Gizmo Warm-up

Burning is an example of a **chemical reaction**. The law of **conservation of matter** states that no atoms are created or destroyed in a chemical reaction. Therefore, a balanced **chemical equation** will show the same number of each type of atom on each side of the equation.

To set up an equation in the *Chemical Equations* Gizmo, type the **chemical formulas** into the text boxes of the Gizmo. First, type in "H<sub>2</sub>+O<sub>2</sub>" in the **Reactants** box and "H<sub>2</sub>O" in the **Products** box. This represents the reaction of hydrogen and oxygen gas to form water.



1. Check that the **Visual** display is chosen on each side of the Gizmo, and count the atoms.

A. How many hydrogen atoms are on the **Reactants** side?

2

**Products** side?

2

B. How many oxygen atoms are on the **Reactants** side?

2

**Products** side?

1

2. Based on what you see, is this equation currently balanced, meaning are there **equal** numbers of each reactant and product on both sides of the equation?

This equation is not currently balanced.

### Activity A:

#### Interpreting chemical formulas

#### Get the Gizmo ready:

- Erase the chemical formulas in each text box.
- Check that the **Visual** displays are selected.



# Teacher Guide Student Exploration Chemical Equations Gizmo

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A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the Speedy Publishing text.

## Teacher Guide Student Exploration Chemical Equations Gizmo:

**Balancing Chemical Equations**, 2009 Inquiries in Science Chemistry Series Balancing Chemical Equations Teacher's Guide *Step By Step Guide - How To Balance Chemical Equations (Quick Review Notes)* E Staff, Step By Step Guide How To Balance Chemical Equations Quick Review Notes Learn and review on the go Use Quick Review Chemistry Notes to help you learn or brush up on the subject quickly You can use the review notes as a reference to understand the subject better and improve your grades Easy to remember facts to help you perform better Perfect study notes for all high school and college students 24 Pages **Secret Formulas** Rebecca Tilley, Carolyn Willard, 1998-08 Designed to convey key math and science concepts these sweet and tasty activities provide real life experiences with chemistry allow students to investigate the properties of substances and build student comprehension of cause and effect central to later understanding of scientific experimentation T G back cover [Chemistry Equations & Answers \(Speedy Study Guide\)](#) Speedy Publishing, 2014-11-26 Chemistry can be one of the more difficult subjects in school mainly because memorization is crucial It can be hard to memorize tons of facts and equations A chemistry equations and answers guide does help students some more than others You can refer to the study guide to determine which formula is suitable to use for the question at hand The more savvy student might consider one or more ways to creatively benefit from the guide For example creating useful tricks to aid you during a challenging test or quiz Chemistry may seem difficult but there are useful study methods **Chemistry Equations & Answers** Speedy Publishing, 2014-08-26 General chemistry inorganic chemistry organic chemistry and biochemistry are all difficult courses requiring much memorization for the student Essentially there is no easy way to learn formulas and facts This is why chemistry classes are such challenges to students even the best ones However a chemistry equations and answers study guide can help the student When used as a quick reference guide it can be used often to determine the formulas needed for various questions The astute student can cleverly devise ways to make the guide useful for test questions or other circumstances requiring one of the many chemistry equations **Students Study Guide** Fredeen, Pearson, 2003-07 **Chemistry Equations & Answers** Pamphlet Master, 2014-10-12 This Chemistry Equations Answers study guide is created by Pamphlet Master for students everywhere This tool has a comprehensive variety of college and graduate school topics subjects which can give you what it takes to achieve success not only in school but beyond Included in the pamphlet are Chemical Formula and Equations What is a Chemical Formula Chemical Formula and Equations Subscripts What Is A Balanced Equation How Do We Balance The Equation What About These Halves Examples of Balancing Chemical Equations **Chemical Principles Student's Study Guide & Solutions Manual** John Krenos, Joseph Potenza, Loretta Jones, Lynn Koplitz, Thomas Spence, 2004-03-19 This combination manual is designed to help students avoid common mistakes and understand the material better The solutions manual section includes detailed answers and explanations to the odd numbered exercises in the text [The Ultimate Chemical Equations Handbook, Teacher Edition](#)

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