

1 Teoria atomica e leggi quantitative

L'Universo è composto di materia ed energia, due aspetti della medesima entità visto che, come dimostrò Einstein, esse possono convertirsi l'una nell'altra secondo la relazione $E = mc^2$.

La chimica è la scienza che studia le caratteristiche, la struttura e le trasformazioni della materia e gli scambi energetici connessi a tali trasformazioni.

Oggi noi possediamo un modello sufficientemente dettagliato per descrivere la struttura della materia, il modello atomico. Parlare di atomi è attualmente quasi un fatto scontato, ma il modello atomico si affermò con difficoltà, nonostante sia stato proposto per la prima volta più di duemila anni fa.

1.1 La struttura della materia: atomi ed elementi

Le prime ipotesi sulla struttura e sulla natura della materia vennero infatti avanzate, agli albori del pensiero occidentale, in Grecia. A tal proposito possiamo individuare, in estrema sintesi, due problemi fondamentali intorno ai quali si sviluppò il dibattito filosofico su questo tema:

A) continuità/discontinuità della materia: l'atomo

Da una parte **Aristotele** che ipotizza l'esistenza di una materia continua, divisibile indefinitamente in frammenti sempre più piccoli e quindi irriducibile ad unità elementari.

due correnti di pensiero che interpretavano la materia in modi diametralmente opposti.

Dall'altra **Democrito di Abdera** (IV sec a.C.), anche se il primo ad avere parlato di atomi fu **Leucippo**, che ritiene invece la materia discontinua e sostiene pertanto l'esistenza di particelle minuscole, chiamate atomi¹, invisibili, incorruttibili ed eterne.

Le idee di Democrito sopravvissero nei secoli. Furono divulgate da **Epicuro di Samo** (III sec. a.C.) e, nel mondo latino, da **Tito Lucrezio Caro** (95-55 a.C.) nel "De Rerum Natura".

B) complessità/semplificità della materia: l'elemento

La ricerca dell'*archè*, del "principio primo", dal quale discendeva tutta la molteplicità dell'essere, rappresentò forse l'elemento peculiare della prima speculazione greca intorno al mondo. Il problema di poter ricondurre l'enorme varietà di sostanze conosciute alla combinazione di poche sostanze semplici significava fondamentalmente razionalizzare il mondo e quindi "spiegare" i fenomeni complessi riconducendoli e riducendoli alla loro composizione elementare.

A parte i primi tentativi effettuati in tal senso dai primi "Fisiologi" presocratici, come **Talete** (624-546 a.C.) che individuò nell'acqua il principio primo o **Anassimene** (586-528 a.C.) che lo identificò nell'aria, l'ipotesi che ebbe la maggior fortuna durante tutto il medioevo, grazie all'autorevole consenso di **Aristotele**, fu senza dubbio quella di **Empedocle** (490-430 a.C.). Secondo tale ipotesi tutta la materia era composta da quattro sostanze fondamentali ed elementari: l'aria, l'acqua, la terra e il fuoco.

Il concetto di atomo e di elemento sono fondamentali in chimica. Le prime definizioni moderne si devono a **Boyle** e a **Newton**.

Per **R. Boyle** (1627-1691) gli elementi sono "corpi primitivi, semplici, incontaminati, che, non essendo costituiti da altri corpi o di loro mescolanze, sono ingredienti di cui i corpi misti sono costituiti e nei quali questi possono essere in definitiva risolti".

Mentre **Newton** immaginò gli atomi come minuscole sfere, dominate solo da forze attrattive e repulsive.

¹dal greco ατομος, indivisibile.

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Richard Bailey



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Manuale della sicurezza dei gas tossici. Guida completa per la preparazione all'esame di abilitazione Calabrese Maurizio, Di Giuseppe Dina, Rossini Renato, 2017-02-28 Uno strumento fondamentale per affrontare l'esame di abilitazione e per gli operatori pubblici e privati che si occupano di sicurezza ed igiene del lavoro relativamente ai gas tossici. Il tema gas tossici pur se datato nella sua impostazione normativa originaria deve essere letto facendo riferimento al progressivo recepimento di norme comunitarie che sono state emanate negli ultimi decenni. Le Direttive comunitarie costituiscono nel campo della sicurezza una vera e propria rivoluzione che pur nelle innovazioni contenute ha lasciato in piedi molti istituti previgenti tra cui per l'appunto quello relativo al R.D. 147/27 sui gas tossici. La prima parte del manuale sostanzialmente di carattere generale con l'illustrazione delle disposizioni del Regio Decreto le novità regolamentari europee sulle sostanze chimiche REACH e CLP, il sistema organizzativo di gestione della sicurezza previsto dal D.Lgs. n. 81/08 e le modalità d'accertamento degli organi di vigilanza. La seconda parte tratta l'imballaggio e stoccaggio, l'impiego dei gas, delle norme riguardanti il trasporto ADR, la prevenzione incendi, la sorveglianza sanitaria e delle misure di primo soccorso, dei dispositivi di protezione individuale, l'applicazione della direttiva ATEX e della direttiva Seveso. Inoltre contiene utilissime schede suddivise per singolo gas tossico organizzate in maniera tale da essere strumento di consultazione e di lavoro immediato per tutti coloro che si occupano di gas tossici. Il CD Rom allegato contiene modulistica normativa, glossario REACH, CLP e ADR.

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