Six Types of Chemical Reaction Worksheet Answers

Balance the following reactions and indicate which of the six types of chemical reaction are being represented:

- a) 2 NaBr + 1 Ca(OH)₂ → 1 CaBr₂ + 2 NaOH
 Type of reaction: double displacement
- b) 2 NH₃+ 1 H₂SO₄ → 1 (NH₄)₂SO₄

 Type of reaction: synthesis
- c) $4 C_9H_9O + 27 O_2 \rightarrow 20 CO_2 + 18 H_2O$ Type of reaction: combustion
- d) 3 Pb + 2 H₃PO₄ → 3 H₂ + 1 Pb₃(PO₄)₂
 Type of reaction: single displacement
- e) 1 Li₃N + 3 NH₄NO₃ → 3 LiNO₃ + 1 (NH₄)₃N

 Type of reaction: double displacement
- f) 3 HBr + 1 Al(OH)₃ \rightarrow 3 H-₂O + 1 AlBr₃ Type of reaction: acid-base
- g) What's the main difference between a double displacement reaction and an acid-base reaction? Acid-base reactions form water.
 - h) Complete combustion reactions always result in the formation of water.
 What other types of chemical reaction may result in the formation of water?
 Acid-base: HCI + NaOH → H₂O + NaCI

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