# Types of Chemical Reactions

Do atoms rearrange in predictable patterns during chemical reactions?

#### Why?

Recognizing patterns allows us to predict future behavior. Weather experts use patterns to predict dangerous storms so people can get their families to safety. Political analysts use patterns to predict election outcomes. Similarly, chemists classify chemical equations according to their patterns to help predict products of unknown but similar chemical reactions.

See B.

#### Model 1 - Types of Reactions

Set A
$4\text{Fe}(s) + 3\text{O}_3(g) \rightarrow 2\text{Fe}_2\text{O}_3(s)$
$N_y(g) + 3H_y(g) \rightarrow 2NH_y(g)$
$2SO_2(g) + O_2(g) \rightarrow 2SO_3(g)$
$MgO(s)+H_2O(l) \rightarrow Mg(OH)_3(aq)$
$P_2O_5(g) + 3H_2O(l) \rightarrow 2H_3PO_4(aq)$
$SO_{3}(g) * H_{2}O(l) \rightarrow H_{2}SO_{4}(aq)$

#### 

With the state of	4
$MgCO_{3}(s) \rightarrow MgO(s) + CO_{2}(g)$	
$8\text{Li}_2S(s) \rightarrow 16\text{Li}(s) + S_a(s)$	
$2H_2O(1) \rightarrow 2H_2(g) + O_2(g)$	
$2KClO_3(s) \rightarrow 2KCl(s) + 3O_2(g)$	

 $2Na_sO_s(s) \rightarrow 2Na_sO(s) + O_s(g)$ 

Na,CO,(aq) + CaCl,(aq) --

"
$$CaCO_3(s) + 2NaCl(aq)$$
  
 $FeS(s) + 2HCl(aq) \rightarrow H_2S(g) + FeCl_3(aq)$   
 $HCl(aq) + NaOH(aq) \rightarrow H_2O(l) + NaCl(aq)$   
 $FeBr_2(aq) + K_2PO_2(aq) \rightarrow FePO_2(s) + 3KBr(aq)$ 

The chemical equations in Model 1 contain the phase notations (s), (f), (g), and (aq). March
each symbol with its meaning.

dissolved in water liquid solid gas
(aq) (L) (S) (9

 Based on the examples provided, which set(s) of reactions in Model 1 typically involve ions in solution (A, B, C or D)?

C aD

 Based on the examples provided, which set(s) of reactions in Model 1 typically involve gases and/or solids?

13

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INGO W. D. HACKH

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teaching from the front and others that can be built into traditional lecture courses to enhance the learning experience

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