

## 1 Teoria atomica e leggi quantitative

L'Universo è composto di materia ed energia, due aspetti della medesima entità visto che, come dimostrò Einstein, esse possono convertirsi l'una nell'altra secondo la relazione  $E = mc^2$ .

La chimica è la scienza che studia le caratteristiche, la struttura e le trasformazioni della materia e gli scambi energetici connessi a tali trasformazioni.

Oggi noi possediamo un modello sufficientemente dettagliato per descrivere la struttura della materia, il modello atomico. Parlare di atomi è attualmente quasi un fatto scontato, ma il modello atomico si affermò con difficoltà, nonostante sia stato proposto per la prima volta più di duemila anni fa.

### 1.1 La struttura della materia: atomi ed elementi

Le prime ipotesi sulla struttura e sulla natura della materia vennero infatti avanzate, agli albori del pensiero occidentale, in Grecia. A tal proposito possiamo individuare, in estrema sintesi, due problemi fondamentali intorno ai quali si sviluppò il dibattito filosofico su questo tema:

#### A) continuità/discontinuità della materia: l'atomo

Da una parte **Aristotele** che ipotizza l'esistenza di una materia continua, divisibile indefinitamente in frammenti sempre più piccoli e quindi irriducibile ad unità elementari.

due correnti di pensiero che interpretavano la materia in modi diametralmente opposti.

Dall'altra **Democrito di Abdera** (IV sec a.C.), anche se il primo ad avere parlato di atomi fu **Leucippo**, che ritiene invece la materia discontinua e sostiene pertanto l'esistenza di particelle minuscole, chiamate atomi<sup>1</sup>, invisibili, incorruttibili ed eterne.

Le idee di Democrito sopravvissero nei secoli. Furono divulgate da **Epicuro di Samo** (III sec. a.C.) e, nel mondo latino, da **Tito Lucrezio Caro** (95-55 a.C.) nel "De Rerum Natura".

#### B) complessità/semplificità della materia: l'elemento

La ricerca dell'*archè*, del "principio primo", dal quale discendeva tutta la molteplicità dell'essere, rappresentò forse l'elemento peculiare della prima speculazione greca intorno al mondo. Il problema di poter ricondurre l'enorme varietà di sostanze conosciute alla combinazione di poche sostanze semplici significava fondamentalmente razionalizzare il mondo e quindi "spiegare" i fenomeni complessi riconducendoli e riducendoli alla loro composizione elementare.

A parte i primi tentativi effettuati in tal senso dai primi "Fisiologi" presocratici, come **Talete** (624-546 a.C.) che individuò nell'acqua il principio primo o **Anassimene** (586-528 a.C.) che lo identificò nell'aria, l'ipotesi che ebbe la maggior fortuna durante tutto il medioevo, grazie all'autorevole consenso di **Aristotele**, fu senza dubbio quella di **Empedocle** (490-430 a.C.). Secondo tale ipotesi tutta la materia era composta da quattro sostanze fondamentali ed elementari: l'aria, l'acqua, la terra e il fuoco.

Il concetto di atomo e di elemento sono fondamentali in chimica. Le prime definizioni moderne si devono a **Boyle** e a **Newton**.

Per **R. Boyle** (1627-1691) gli elementi sono "corpi primitivi, semplici, incontaminati, che, non essendo costituiti da altri corpi o di loro mescolanze, sono ingredienti di cui i corpi misti sono costituiti e nei quali questi possono essere in definitiva risolti".

Mentre **Newton** immaginò gli atomi come minuscole sfere, dominate solo da forze attrattive e repulsive.

<sup>1</sup>dal greco ατομος, indivisibile.

# Sommario Sommario 1 Teoria Atomica E Leggi Quantitative 1 1

**Wenbin Ji**



## **Sommario Sommario 1 Teoria Atomica E Leggi Quantitative 1 1:**

Manuale della sicurezza dei gas tossici. Guida completa per la preparazione all'esame di abilitazione Calabrese Maurizio, Di Giuseppe Dina, Rossini Renato, 2017-02-28 Uno strumento fondamentale per affrontare l'esame di abilitazione e per gli operatori pubblici e privati che si occupano di sicurezza ed igiene del lavoro relativamente ai gas tossici. Il tema gas tossici pur se datato nella sua impostazione normativa originaria deve essere letto facendo riferimento al progressivo recepimento di norme comunitarie che sono state emanate negli ultimi decenni. Le Direttive comunitarie costituiscono nel campo della sicurezza una vera e propria rivoluzione che pur nelle innovazioni contenute ha lasciato in piedi molti istituti previgenti tra cui per l'appunto quello relativo al R D 147/27 sui gas tossici. La prima parte del manuale sostanzialmente di carattere generale con l'illustrazione delle disposizioni del Regio Decreto le novità regolamentari europee sulle sostanze chimiche REACH e CLP, il sistema organizzativo di gestione della sicurezza previsto dal D Lgs n. 81/08 e le modalità d'accertamento degli organi di vigilanza. La seconda parte tratta l'imballaggio e stoccaggio, l'impiego dei gas, delle norme riguardanti il trasporto ADR, la prevenzione incendi, la sorveglianza sanitaria e delle misure di primo soccorso, dei dispositivi di protezione individuale, l'applicazione della direttiva ATEX e della direttiva Seveso. Inoltre contiene utilissime schede suddivise per singolo gas tossico organizzate in maniera tale da essere strumento di consultazione e di lavoro immediato per tutti coloro che si occupano di gas tossici. Il CD Rom allegato contiene modulistica normativa, glossario REACH, CLP e ADR.

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## **Table of Contents Sommario Sommario 1 Teoria Atomica E Leggi Quantitative 1 1**

1. Understanding the eBook Sommario Sommario 1 Teoria Atomica E Leggi Quantitative 1 1
  - The Rise of Digital Reading Sommario Sommario 1 Teoria Atomica E Leggi Quantitative 1 1
  - Advantages of eBooks Over Traditional Books
2. Identifying Sommario Sommario 1 Teoria Atomica E Leggi Quantitative 1 1
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Sommario Sommario 1 Teoria Atomica E Leggi Quantitative 1 1
  - User-Friendly Interface
4. Exploring eBook Recommendations from Sommario Sommario 1 Teoria Atomica E Leggi Quantitative 1 1
  - Personalized Recommendations
  - Sommario Sommario 1 Teoria Atomica E Leggi Quantitative 1 1 User Reviews and Ratings
  - Sommario Sommario 1 Teoria Atomica E Leggi Quantitative 1 1 and Bestseller Lists
5. Accessing Sommario Sommario 1 Teoria Atomica E Leggi Quantitative 1 1 Free and Paid eBooks
  - Sommario Sommario 1 Teoria Atomica E Leggi Quantitative 1 1 Public Domain eBooks
  - Sommario Sommario 1 Teoria Atomica E Leggi Quantitative 1 1 eBook Subscription Services
  - Sommario Sommario 1 Teoria Atomica E Leggi Quantitative 1 1 Budget-Friendly Options

6. Navigating Sommario Sommario 1 Teoria Atomica E Leggi Quantitative 1 1 eBook Formats
  - ePub, PDF, MOBI, and More
  - Sommario Sommario 1 Teoria Atomica E Leggi Quantitative 1 1 Compatibility with Devices
  - Sommario Sommario 1 Teoria Atomica E Leggi Quantitative 1 1 Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Sommario Sommario 1 Teoria Atomica E Leggi Quantitative 1 1
  - Highlighting and Note-Taking Sommario Sommario 1 Teoria Atomica E Leggi Quantitative 1 1
  - Interactive Elements Sommario Sommario 1 Teoria Atomica E Leggi Quantitative 1 1
8. Staying Engaged with Sommario Sommario 1 Teoria Atomica E Leggi Quantitative 1 1
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Sommario Sommario 1 Teoria Atomica E Leggi Quantitative 1 1
9. Balancing eBooks and Physical Books Sommario Sommario 1 Teoria Atomica E Leggi Quantitative 1 1
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Sommario Sommario 1 Teoria Atomica E Leggi Quantitative 1 1
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Sommario Sommario 1 Teoria Atomica E Leggi Quantitative 1 1
  - Setting Reading Goals Sommario Sommario 1 Teoria Atomica E Leggi Quantitative 1 1
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Sommario Sommario 1 Teoria Atomica E Leggi Quantitative 1 1
  - Fact-Checking eBook Content of Sommario Sommario 1 Teoria Atomica E Leggi Quantitative 1 1
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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