chapter 9 Alcohols, Ethers, and Related Compounds

Chapter Review

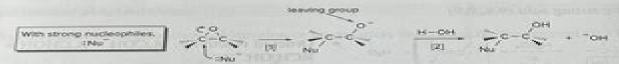
General facts about ROH, ROR, and epoxides

All three compounds contain an O atom that is sp' hybridized and tetrahedral (9.2).

 All three compounds have polar C=O bonds, but only alcohols have an O=H bond for intermolecular hydrogen bonding (9.4).

Alcohols and others do not contain a good leaving group. Nucleophilic substitution can occur only
after the OH (or OR) group is converted to a better leaving group (9.7A).

 Epoxides have a leaving group located in a strained three-membered ring, making them reactive to strong nucleophiles and acids HZ that contain a nucleophilic atom Z (9.16).



A new reaction of carbocations (9.9)

 Less stable carbocations rearrange to more stable carbocations by shift of a hydrogen atom or an alkyl group. Besides rearrangement, carbocations also react with nucleophiles (7.12) and bases (8.6).

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Table of Contents Solutions Manual Organic Chemistry 6th Edition

- 1. Understanding the eBook Solutions Manual Organic Chemistry 6th Edition
 - The Rise of Digital Reading Solutions Manual Organic Chemistry 6th Edition
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Solutions Manual Organic Chemistry 6th Edition
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Solutions Manual Organic Chemistry 6th Edition
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Solutions Manual Organic Chemistry 6th Edition
 - Personalized Recommendations

- Solutions Manual Organic Chemistry 6th Edition User Reviews and Ratings
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 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Solutions Manual Organic Chemistry 6th Edition
 - Setting Reading Goals Solutions Manual Organic Chemistry 6th Edition
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Solutions Manual Organic Chemistry 6th Edition
 - Fact-Checking eBook Content of Solutions Manual Organic Chemistry 6th Edition

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- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
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