

Time : 2 Hrs.

Max. Marks : 40

Q.1 Select and write the most appropriate answer from the given alternatives for each sub-question. [8]

- (i) Which of the following is an intensive property ? (1)
 (a) Internal energy (b) Heat capacity
 (c) Enthalpy (d) Surface tension.
- (ii) The species which acts as an conjugate acid as well as base is (1)
 (a) HSO_4^- (b) H_2SO_4
 (c) NH_4OH (d) CO_3^{2-}
- (iii) The electrolyte used in Hydrogen - Oxygen fuel cell is (1)
 (a) ZnSO_4 (b) NaOH
 (c) 20% H_2SO_4 (d) KCl
- (iv) During α - decay, mass number decreases by how many units ? (1)
 (a) 2 (b) 3
 (c) 4 (d) 5
- (v) For the first order reaction, the half-life period is dependent on (1)
 (a) catalyst (b) temperature
 (c) square root of initial concentration of reactants
 (d) rate constant
- (vi) Which of the following element does NOT exhibit variable oxidation states ? (1)
 (a) Iron (b) Copper
 (c) Zinc (d) Manganese
- (vii) The difference between heat of combustion of liquid benzene at constant pressure and at constant volume at 300 K is (1)
 ($R = 8.314 \text{ J/K/mol}$)
 (a) -3741 J (b) 3741 J
 (c) 37.41 kJ (d) -37.41 kJ
- (viii) The normality of solution containing $4 \times 10^{-3} \text{ kg}$ of NaOH dissolved in 0.5 m^3 of solution is (1)
 (At. wt. of $\text{Na} = 23, \text{O} = 16, \text{H} = 1$)
 (a) 0.2 N (b) 0.02 N
 (c) 0.002 N (d) 0.0002 N

Q.2 (A) Attempt any One : [8]

- (i) State and explain van't Hoff - Charles' law. (2)
 (ii) Distinguish between isotopes and isobars. (2)

(B) Attempt any One :

- (i) State Rate law. Write 'one' example of it. (2)
 (ii) Why do the d-block elements form coloured compounds ? (2)

(C) Answer the following :

- (i) Define : (a) Isotonic solutions, (b) Nuclear fission. (2)
 (ii) Give the limitations of Arrhenius theory of acids and bases. (2)

Q.3 (A) Attempt any One : [8]

- (i) What is Pseudo-first order reaction ? (3)
 Distinguish between molecularity of reaction and order of reaction. (3)
 (ii) What is common ion effect ? Explain its application in precipitation of Group II radicals. (3)

(B) Attempt any One :

- (i) Explain the reduction of zinc oxide to zinc by modern vertical retort process. (3)
 (ii) Explain the measurement of e.m.f. of a cell using potentiometer. (3)

(C) Answer the following :

Prove that $\Delta H = \Delta E + \Delta nRT$ (2)

Solved Exercises Of Chemistry Of 11th Hsc

Rachel S Tattersall



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