

PAPER-2 [Code – 5]
IITJEE 2011
PART - I: CHEMISTRY

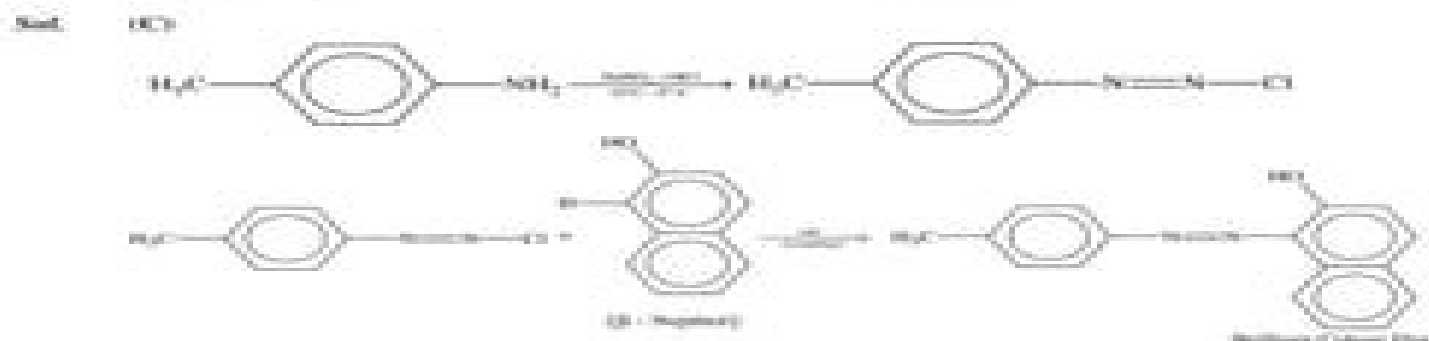
SECTION – I (Total Marks : 24)
(Single Correct Answer Type)

This Section contains 8 multiple choice questions. Each question has four choices (A), (B), (C) and (D) out of which **ONLY ONE** is correct.

1. The freezing point (in °C) of a solution containing 0.1 g of $K_3[Fe(CN)_6]$ (Mol. Wt. 329) in 100 g of water ($K_f = 1.86 \text{ K kg mol}^{-1}$) is
 (A) -2.3×10^{-1} (B) -5.7×10^{-1}
 (C) -5.7×10^{-2} (D) -1.2×10^{-1}

Sol. (A)
 $K_3[Fe(CN)_6] \rightarrow 3K^+ + [Fe(CN)_6]^{4-}$
 $i = 4$
 $\Delta T_f = K_f \times i \times \frac{m}{M} \times \frac{1000}{W} = 1.86 \times 4 \times \frac{0.1}{329} \times \frac{1000}{100} = 2.3 \times 10^{-1}$
 $T_f = -2.3 \times 10^{-1}$

2. Amongst the compounds given, the one that would form a brilliant colored dye on treatment with $NaNO_2$ in dil. HCl followed by addition to an alkaline solution of β -naphthol is



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BS Sijwalii,Ajay Singh



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