

Name \_\_\_\_\_ Date \_\_\_\_\_ Class \_\_\_\_\_

## CHAPTER 15 STUDY GUIDE FOR CONTENT MASTERY

### Solutions

#### Section 15.1 What are solutions?

In your textbook, read about the characteristics of solutions.

Use each of the terms below just once to complete the passage.

immiscible	liquid	solution	solute	solution
insoluble	miscible	solute	solvent	

Air is air (1) solution of oxygen gas dissolved in nitrogen gas. The oxygen in air is the (2) solute and nitrogen is the (3) solvent. Because oxygen gas dissolves in a solvent, oxygen gas is air (4) soluble substances. A substance that does not dissolve is (5) insoluble. (6) Liquid solutions are the most common type of solutions. If one liquid is soluble in another liquid, such as acetic acid in water, the two liquids are (7) miscible. However, if one liquid is insoluble in another, the liquids are (8) immiscible.

Read about solvation in aqueous solutions in your textbook.

The diagram shows the hydration of solid sodium chloride to form an aqueous solution. Use the diagram to answer the following questions.



8. Hydration is solvation in which the solvent is water. What is solvation?

Solution is the process of surrounding solute particles with solvent particles to form a solution.

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## CHAPTER 15 STUDY GUIDE FOR CONTENT MASTERY

#### Section 15.1 continued

10. As sodium chloride dissolves in water, what happens to the sodium and chloride ions?

The sodium and chloride ions are separated and surrounded by the water molecules.

11. Explain the attraction of the water molecules around the sodium ions and chloride ions.

Because the sodium ion is positively charged, it attracts the negatively charged portion of the water molecule (the oxygen atom) and repels the positively charged portion of the water molecule (the hydrogen atoms). Because the chloride ion is negatively charged, it attracts the positively charged portion of the water molecule and repels the negatively charged portion.

12. How does the strength of the attraction between water molecules and sodium and chloride ions compare with the strength of the attraction between the sodium ions and chloride ions? How do you know?

The attraction between the water molecules and the sodium and chloride ions is greater than the attraction between the sodium and chloride ions. The greater strength of attraction between the water molecules and the ions is what causes the solvation process to occur.

13. List three ways that the rate of solvation may be increased.

stirring or shaking the solution, breaking the solute into smaller pieces, and heating the solvent

In your textbook, read about heat of solution, solubility, and factors that affect solubility.

For each statement below, write true or false.

True \_\_\_\_\_ 14. The overall energy change that occurs when a solution forms is called the heat of solution.

False \_\_\_\_\_ 15. Solubility is a measure of the maximum amount of solute that dissolves in a given amount of solvent at a specified temperature and pressure.

False \_\_\_\_\_ 16. Solvation continues as long as the solvation rate is less than the crystallization rate.

True \_\_\_\_\_ 17. In a saturated solution, solvation and crystallization are in equilibrium.

True \_\_\_\_\_ 18. Additional solute can be dissolved in an unsaturated solution.

True \_\_\_\_\_ 19. The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

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**Maheshkar, Chandan, Sharma, Vinod**



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Vinod,2018-06-08 Due to the impact of globalization business practices have been constantly evolving throughout the twenty first century Teachers and instructors are faced with the challenges of educating future business leaders on evolving concepts such as diversity cultural environments and the integration of countries and economies Through these challenges it is critical for educators to understand teaching tools and pedagogical practices that bring this globalized outlook into the classroom The Handbook of Research on Cross Cultural Business Education is an important research publication that explores the role of teaching tools used in business education to increase competitive business skills and make them applicable in cross cultural business environments It also provides business and management educators with teaching practices that promote learning via exposure to global cultures Featuring a broad array of topics such as pedagogical modeling cross cultural learning and social responsibility this is an important resource for business and management educators business managers policy makers researchers business professionals and graduate students **Hybrid Learning**

Philip Tsang,Simon K.S. Cheung,Victor S.K. Lee,Ronghuai Huang,2010-08-12 The Third International Conference on Hybrid Learning ICHL 2010 was organized by the School of Continuing and Professional Studies of The Chinese University of Hong Kong Beijing Normal University Goethe Institut China Caritas Francis Hsu College and Caritas Bianchi College of Careers ICHL 2010 provided a platform for knowledge exchange on hybrid learning among educators researchers and computer

scientists who share a common goal to enhance the quality of learning and teaching in this fast changing knowledge world with the support of technology innovation Hybrid learning has been an ongoing trend for a number of years It is not merely a simple combination of face to face and technology mediated instruction but also encompasses different learning strategies for teaching and learning It places emphasis on outcome based teaching and learning and provides a diversified learning environment Hybrid learning is probably the most efficient learning mode in the present age of globalization when learning has to be borderless and dynamic in order to address the diverse learning needs of students Students are given more opportunities to develop into active independent learners and to practice practical skills for work and study It was our pleasure to have three keynote speakers for the ICHL 2010 They were Rebecca Launer from Goethe Institut Germany Bebo White from Stanford University and Yan Ji Chang from Tsinghua University all of whom shared with us their valuable insights in the hybrid learning field

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*Electrotechnology Practice* Steven Hanssen, Jeffery Hampson, 2023-03-28

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**Content-Area Literacy** Tom Bean, 2011-09. Use a fresh 21st-century skills approach to address the common difficulties associated with teaching adolescents to read content-area material. The strategies presented in this book will allow teachers to differentiate instruction to best meet students' literacy needs. This resource is aligned to College and Career Readiness Standards. *Ebook: Chemistry* Julia Burdge, 2014-10-16. Chemistry Third Edition by Julia Burdge offers a clear writing style written with the students in mind. Julia uses her background of teaching hundreds of general chemistry students per year and creates content to offer more detailed explanation on areas where she knows they have problems. With outstanding art, a consistent problem-solving approach, interesting applications woven throughout the chapters, and a wide range of end-of-chapter problems, this is a great third-edition text.

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